

Ground Water Resources of Jalore District Part-II Hydrochemistry

GOVERNMENT OF RAJASTHAN

# SURVEY & RESEARCH

## **GROUND WATER DEPARTMENT**

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GROUND WATER RESOURCES OF JALORE DISTRICT PART - II HYDROCHEMISTRY

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OFFICE OF THE SENIOR HYDROGEOLOGIST GROUND WATER DEPARTMENT (D.P.A.P.) PALI

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## GROUND WATER RESOURCES OF JALORE DISTRICT, RAJASTHAN

PART - II HYDROCHEMISTRY 

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Ground Water Resources of Jalore District Part-II Hydrochemistry

#### ABSTRACT

The report deals with the hydrochemical investigations carried out between 1976 - 1982 in Jalore district, as a part of detailed ground water surveys under Drought Prone Area Programme (DPAP). During these surveys 9,692 water samples were collected periodically and repeatedly, from various water bearing formations and were subjected to chemical analysis. They represent formations ranging from crystalline to the quaternary.

Ground water in the district is fresh to saline and the electrical conductivity in phreatic aquifer and even in deep aquifer varies widely. It is between 430 ALS/CM to 27,000 ALS/CM and 360 ALS/CM to 50,000 ALS/CM in phreatic and deeper aquifer respectively. Jaswantpura and Raniwara Panchayat Samities have mostly fresh water, while saline ground water mostly occurs in Sanchore, north of Bhinmal, west of Saila, north of Ahore and Jalore.

Ground water occuring in younger alluvium except in west of Sanchore, is fresh to moderately saline having  $HCO_3 - Cl$  type to  $Cl - HCO_3$  type of chemical character, while in older alluvium and tertiary it is mineralised with, predominantly, sodium salts. Waters representing granitic formation are mostly potable and soft with high fluoride content. Nitrate in water is observed in all the formations but it is comparatively less in deeper aquifer.

Bicarbonate type of water occurs along hilly terrain and foot hill zones of Jaswantpura, Raniwara and extreme northern part of Saila close to Siwana hills, which from the recharge zone. The ground water changes to mixed type around Bhinmal, Saila and Jalore to finally chloride type in west of Sanchore, north of Bhinmal, north-west of Saila and north of Ahore. - ii -

Ground water from phreatic and deeper aquifers of Jaswantpura and Raniwara is suitable for general domestic use. They are mostly free from harm-ful level of nitrate and fluoride. In parts of Ahore, Jalore, Saila and Bhinmal blocks waters are tolerable but found to contain high fluoride as evident from dental fluorosis amongst the local population. However, drinking water is a problem in Sanchore, north of Bhinmal and west of Saila where it is saline and in some area high nitrate also probilits its use for drinking.

Water quality evaluation for agriculture based on water rating Criteria (1972), has indicated that ground water upto 8,000 MS/CM is suited for use in irrigation. Soils conditions are favourable for growing salt tolerant crops. Few better quality area show high RSC and sodium percentage in ground water, where application of gypsum to the soils before irrigation is suggested.

#### CHAPTER - 1

#### INTRODUCTION

Ground water investigations in part of Jalore district were carried out from time to time by State and Central Ground Water Departments in the last two decades. Based on the findings of these investigations, various minor irrigation schemes and other programmes for exploitation of ground water were subsequently formulated and implemented.

Because of these ground water development programmes certain fresh water areas appeared to be over exploited while vast area of saline track remained untapped, due to nonavailability of detailed informations on hydrogeological and hydrochemical characters of the aquifers.

Therefore, detailed hydrogeological investigation under Drought Prone Area Programme (D.P.A.P.) were initiated from the year 1976 - 77 to evaluate the ground water potential and to delineate precisely the fresh water aquifers in saline area of the district.

The hydrochemical investigations formed a part of these studies with the objective to study the distribution of various dissolved salts in ground water, their relationship to aquifer characteristics and their suitability for drinking, domestic agricultural and livestocks purposes.

#### LOCATION:

Jalore district is situated in the south western part of Rajasthan and has an areal extent of 10,640 Sq.Kms. It lies between latitude  $24^{\circ}31'$  to  $25^{\circ}45'$  and longtitude  $71^{\circ}7'$  to  $73^{\circ}6'$ .

#### TOPOGRAPHY

In Jalore district elevation of the area ranges from 76.20 mts. above M.S.L. in the ewest, at the confluence of the Luni and Jawai - Sukari rivers, to 609.60 metres above M.S.L. in the east. Generally, the terrain slopes towards west, the valley floor has an average elevation ranging from 60.96 metres to 213.36 metres above M.S.L. The hill tops are normally shaped by weathering which has given rise to tors and boulders of various shapes and sizes in the eastern part while sand dunes are common topographic features in mid-eastern and western part.

Geomorphologically, the alluvial valley floor belongs to a natural landslope system. Much of the drainage patterns was established during plei--stocene and sub-recent times. The arid cycle thus conditioned the initial fluvial land forms through the formation of sand dunes and sandy plains, super imposed on earlier aggregated alluvial plains and slopes.

#### DRAINAGE - CLIMATE - RAINFALL

The region is drained by the Luni river and its tributaries originating from the Aravalli hill ranges. The main tributaries of Luni river are Jawai, Sukari, Khari, Bandi and Sagi. All rivers are ephemeral with braided meandering courses and

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wide flood plains. Besides these, there are innumerable old channels buried under wind blown sand.

The area lies in the arid to semi arid zone of Rejasthan with extremes of climate and high diurnel variations in temperature. May - June is the hottest part of the year when day temperature rises even upto 43°C while December - January experienced extreme winters with temperature falling as low as 2.5°C.

The rainfall in the region is very low and erratic, the normal annual precipitation record--ed is 384.0 mm (1901 - 1981), which shows the probable error of 188.2 mm. South-west monsoon contributes 92.2% of annual rain fall during monsoon period. Most of the rainfall is lost by evaporation and partly by run off. An average annual rainfall recorded at Jalore Station is 379.86 mm.

#### HYDROGEOLOGY

Geological formations encountered in the area range in age from post Delhi intrusives to the quaternary comprising of loosely consolidated to unconsolidated alluvial deposits and blown sand.

The younger alluvium forms the major promi--sing aquifer in the area having generally fresh water. It is generally loosely consolidated to unconsolidated stream laid deposits of gravels and sands. Older alluvium generally contains saline water or water of mixed character, and is composed of unconsolidated to semi-consolidated sand and gravel with clay, kenkar and silts. - 4 -

Tertiary formations comprising thick consolidated clays, silts and gravel, medium to coarse sand with shale fragment and little clays are found in deeper horizons. It generally yields

Malani volcanics consist of rhyplites and felsites, intercalated with acid tuff in pyrocla astic material. These are having well developed joints system which are capable of giving moderate discharge, generally yielding fresh water.

Jalore granites also have well developed joints and the weathered mantle which yield mainly fresh and mixed type water depending upon topography of the area.

Erinpura granites are weathered, fractured and well jointed, confined to south of Bhinmal. It contains mainly fresh to tolerable quality of water.

#### METHODOLOGY.

The following methodology was adopted to study the hydrochemistry of ground water:-1. On the onset of programme all the earlier water analyses data were compiled and base maps were prepared.

2. In all 9692 water samples were collected between 1976 to 1982. Data of 165 analyses have been selected as representing the different hydrochemical characteristics in the area. These were selected from analysis results of samples collected during the year 1979. The analyses include the determination of major cations and anions namely sodium, potassium, calcium, magnesium, cerbon--ate, bicarbonate, chloride, sulphate, nitrete and fluoride. Specific conductance (expressed in microsiemens per centimeter at 25°C) and pH were also determined instrumentally. The important parameters like total hardness, sodium adsorption ratio, residual sodium carbonate and sodium percentage were calculated for evaluating the suitability for irrigation purpose. These are reported in appendix-I and II. The analyses results are expressed in tilligram per litre and equivalent per litre.

3. Seasonal samplings of water samples from 165 key wells were carried out during May - June (fre Monsoon) Oct. - Nov. (Post monsoon) and March (fost / irrigation) from 1978 to 81.

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4. The data were further projected for thear geochemical types.

5. Chemical data so collected were used for preparation of verious chemical maps and studies for their seasonal variations, suitability for drinking, domestic and irrigation purposes.

6. Forty nine water samples from deeper aquifers representing different water bearing formations, were also studied, in detail, as enumerated in column 4 and 5.

#### ACKNOW IEDGEMENTS .

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## - 7 -<u>CHAPTER - 2</u> <u>CUALITY OF PHREATIC WATER</u>

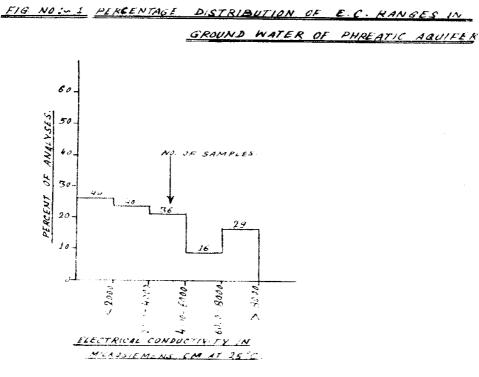
#### EIECTRICAL CONDUCTIVITY:

In arid and semi arid region ground water is often mineralized, Salinity, a natural phenonmenon, occuring in both soils & waters, is dependent upon olimatic conditions of the region. Jalore, being a part of arid region, has ground water with varing degrees of selinity.

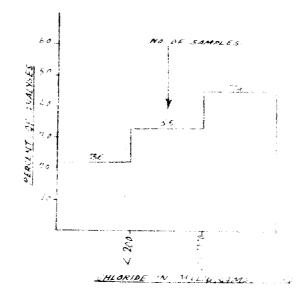
The lowest value of electrical conductivity i.e. 430 microsiemens-cm at 25° is observed in two well waters viz. FANSERI AND RAMPURA of P.S. Jaswantpura and Raniwara respectively. Both the wells are situated near the hilly terrain. The highest value of 27,000 microsiemens/cm at 25°C is recorded in well water of village BAORI of P.S. Abore. The percentage of water samples, in various electrical conductivity ranges of phreatic aquifer, are shown in Fig.1, which indicates the saline character of ground water. Only 26.7% of samples have E.C. value less than 2009 microsiemens.

A map, Plate No.1, showing the electrical conductivity in ground water of district, is prepared on the basis of 165 analysis data. It depicts that fresh ground water (E.C. less than 2000 MS) is found in parts of Jaswantpura, Raniwara and near the river banks around Bhinmal and Saila. Ground water of tolerable quality with 2000 - 4000 E.C. value occurs in western part of Jalore, Ahore and Bhinmal, while Sanchore, north of Ahore, south - west and north east of Jalore and part of Bhinmal have saline waters.

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ELG NO. 2 PERCLATAGE DISTRIBUTION OF CHLORIDE RADIUS IN GROUND WATER OF PHREATIC AQUINA



Since electrical conducitivity of the water is a measure of its total soluble salts, Plate - 2 showing the distribution of T.D.C. is similar to Plate-1. The T.D.S. map Plate - 2 on perusal shows that ground water is gradually getting mineralized from south and south-east to north, north-west and west.

Fresh water mostly contains predominantly bicarbonates of calcium and magnesium but in few well waters sodium-bicarbonate dominates. Analyses showing the tolerable quality of water, have mixed type of solutes while seline water principally contains sodium chloride besides sodium sulphate.

#### CHLORIDE CONTENT:

Chloride is the principal anion in ground water of arid zones. Chloride, which initially comes from rain, gets gradually enriched due to dissolution of the salts from the earths crust through which the ground water circulates.

Chloride content in ground water varies from 17 mg/l (Panseri, P.S. Jaswantpura) to 9112 mg/l (Baori, P.S. Ahore) Fig.2 shows the percentage of water samples in various ranges of chloride content in ground water. In 21.8 percent of analyses the chloride ranges between 17 to 200 mg/l while 33.3 percent of analyses shows its value between 200 to 1000 mg/l indicating the mineralised nature of ground water. Many analyses (44.9%) show chloride content above 1000 mg/l impairing their potability. Fig. 3

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a graph of chloride versus E.C. indicates that the salinity of ground water increases with the increase in concentration of chloride.

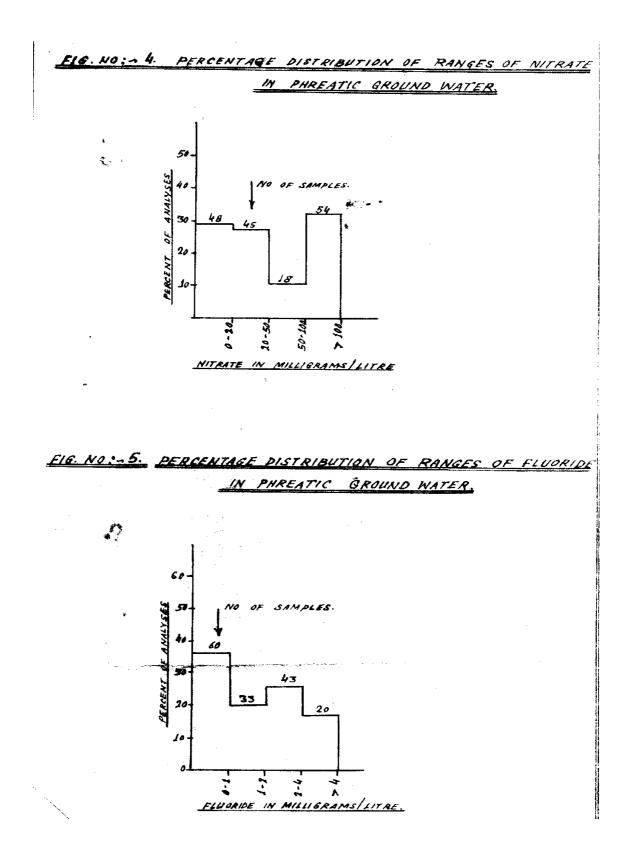
Plate III shows the distribution of chloride of phreatic ground water. It depicts that the chloride in water increases with increasing total dissolved solids from south and south-west to west, north-west and north. Central part covering north of Bhinmal also have high chloride content, exceeding above 1,000 mg/l at some places. High contents of chloride in the area indicate that ground water movement is sluggish and slow.

#### NITRATE:

Nitrate in water is mainly contributed either through sewage pollution or by return flow of irrigation water, leaching nitrogenous fertilizer. However significant amount of nitrate is observed to be contributed by legaume crops to the soils.

Figure-4 shows the different ranges of nitrate in phreatic aquifer and further indicates that 32.7 percent of analyses have nitrate above 100 mg/1. It ranges between zero (many analyses) to 600/1 (water of village Murtara silli, P.S. Jaswantpura). Analyses showing high nitrate content are generally observed to have high contents of potassium which shows the direct effects of fertilizer and return flow of irrigation water polluting the ground water and making it unfit for human consumption.

It is observed from the map showing distribution (plate -2) of nitrate that in northern part of the district covering villages Sobrawas, Nandia,



Jesawas around Saila and in eastern part covering Mera - Uparla, Mayla was around Jalore phreatic aquifer has nitrate content more than 250 mg/l. It is further seen that saline waters of west and north also have nitrate content of more than 100 mg/l. It is difficult to find nitrate free water in the above referred area.

However in the southern and central part of the district ground water is free from harmful level of nitrate.

#### FLUORIDE:

Fluoride is a minor constitutent of igneous and sedimentary rocks. In Jalore district granites and gneissess are the main sources of fluoride for ground water which is evident from the fact that high content of fluoride are found in those wells tapping either granite or gneissess.

Fluoride content in ground water varies from nil to 14.20 mg/l (Morsim, p.s. Bhinmal). The ranges of fluoride and their percentage are shown in figure 5 which depicts that fluoride content exceeded 2 mg/l in 36.36 percent of analyses, while 26.0 percent of analyses falls in the rage of 2 - 4 mg/l of fluoride concentration. The high content of fluoride above 4 mg/l is observed in 17.6 percent of analyses. This shows that fluoride is a proble--matic ingredient in ground water of the district.

It is seen from the plate III that the fluoride contents above 4 mg/l are found in ground water of Ahore, Jalore where wells are tapping granite. Some localised patches around Bhinmal, north-west of Saila, south of Sanchore etc. have fluoride content

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more than 4 mg/l in phreatic aquifer.

There is no relationship between fluoride and salinity. However, close relationship could be seen between R.S.C. and fluoride content of the water.

#### CHEMICAL QUALITY IN RELATION TO AQUIFER:

It is common knowledge that the lithological or minerological character of water bear--ing materials determines to a large extent the chemical quality of ground water. The principal water bearing formation of Jalore district are (a) Younger alluvium (b) Older alluvium (c) Granite (d) Rhyolite. It is observed from the analyses that minimum variation in chemical character is noticed in the granite and rhyolite formations and maximum in alluvial formation. The chemical quality of ground water of the different aquifers is summarised in the table-I.

It is seen from the table - III that the quality of water of these formations is saline except in rhyolite where average range of electrical conductivity of ground water has been found to be 2,450 microsiemens/cm at 25°C.

#### GROUND WATER IN YOUNGER ALLUVIUM:

Fresh to saline water are found in this formation. Fresh water, mostly containing bicarbonate salts, is found along the banks of Jawai - Sukri river. This formation is mostly consisting of coarse to medium sand, gravels, and pebbles which facilitate the quick movement of water due to high permeability thus reducing the contact time of water with the formation materials.

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This aguifer is spread around the confluance of Luni and Jawai - Sukri river, east to west covering southern portion of Raniwara and west of Sanchore i.e. upto Runn. It is represented by 28.48% of total water samples analysed. The range of electrical conductivity in water varies from less than 1,000 to more than 15,000 microsiemens/cm. About 29 % of water samples within this formation range between 5,000 to more than 15,000 microsiemens/cm with an average value of 3,960 microsiemens/cm. The well waters analysed around Sanchore have shown higher value of electrical conductivity i.e. containing more solutes. The reasons being that it is a discharge zone and is situated at lower level where ground water movement is very sluggish. There is enough time for solutes to dissolve more and more salts in water. For this reason the average value of E.C. is higher. The chloride, nitrate and fluoride ranges are varying between 20 - 6717 mg/1, 0-350 mg/1 and 0 - 14.20 mg/1 with everage values of 783 mg/l, 88 mg/l and 2.47 mg/l respectively. Nearly 30% of water samples have chlo--ride content above 1,000 mg/l, nitrate content above 100 mg/l, fluoride content above 2 mg/l and hardness above 600 mgCa-CO3 /I, thus the potable ground water in this aquifer is of tolerable quality.

In majority of cases, the character of ground water is bicarbonate, bicarbonate- chloride and chloridebicarbonate type. This type of chemical behaviour has been observed in ground water around Jawai - Sukri river and around Raniwara. But in discharge zone (west of Sanchore), sodium chloride type of water occurs.

LATIC AQUIFURS.	C C	750 nil 14 350 nil 14 Ankhol Bhawatra Khejdiyal erage Suntri 2.4	600 nil L <u>Nimbla</u> Fadar rage. <u>1.8</u> avei	600 U Murtara- silli.	140 Chatwar rage.
- 13 - TABLE-1: RANGES OF SOME CHEMICAL CONTENTS IN GROUND WATER OF DIFFERENT PHREATIC AQUIERS	Chloride cont- -ent in mg/l. Mini. Max.	bol Alri	UL I	2553 nil lore Rundmal Bhin ki dhani 755 average.	n11 - <u>Iakha</u> rra.
HEMICAL CONTENTS IN GRO	% of Electrical con- total <u>-ductivity in MS</u> water Mini. Max. samp- les.	. 5. 6. 430 19200 <u>Rampure. Ankhol</u> . 3960 average.	430 27000 Panseri Baori 4967 average.	750 Jalore Ru <u>3593</u> averaf	830 5300 <u>Lakhawas Chatwara</u> 2450 average.
BIE-1: RANGES OF SOME C	Aquifer. No.of % water to sample wa each sa aquifer. le	2. 3. 4 Younger alluvium. 47 28.48	<b>2.01der</b> 95 57.58 afluvium.	. Granite 20 12.12	4. Rhyolite 3 1.82

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#### GROUND WATER IN OLDER ALLUVIUM:

This formation is mostly consisting of clay and kankar having low transmisivity. Therefore, the water remains in contact with this formation for longer period as compared to younger alluvium. This phenomenon increases the selinity in ground water. Besides this, the base exchange, a characteristic of clayey formation, also adds to salinity of water as well change the chemical composition of ground water.

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This aquifer is widely distributed in the Jalore district; mainly in north eastern, southern, western and central parts. It is represented by 57.58% of total water samples analysed. The salinity in water varies from 430 to 27,000 microsiemens/cm with an average value of 4,967 microsiemens/cm. Table-III. The chloride, nitrate and fluoride of ground water in this aquifer average to 1,328 mg/l, 109 mg/l. and 1.85 mg/l respectively. It is also observed from the chemical analyses that 37%, 47%, and 32% of water samples have chloride content above 1,000 mg/l, nitrate content above 100 mg/l and Fluoride content above 2 mg/l respectively. If indicates that the availability of potable ground water is limited in this formation.

Practically sodium chloride type of character is exhibited by the ground water of this aquifer except around south-west and north west of Saila where bicarbonate- chloride type or chloride - bicarbonate type waters occur.

GROUND WATER IN GRANITE:

Jalore granites are prophyritic in texture, highly fractured and weathered. These rocks have no primary porosity. The ground water movement takes place through weathered and fractured portions.

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Grainite occurs as hydrogeological formation in north-east, central and south-eastern porition of the district near village Rama, Jalore, Jaswantpura and Bhinmal section respectively. It is represented by 12.12 % of total analyses.

It is observed from the analyses that 45 % well waters representing the formation are saline. The reason being the water movement is restricted due to limited extention of fracture with in the aquifer. Analyses of remaining 55 % well waters represent the foot-hill zones and show fresh to mix type of chemical character. It is mainly due to the extention of joints, fractures and weathered mantle in foot hill zones where water movement is quick. The chloride, nitrate and fluoride contents of the water in this aquifer average to 755 mg/1, 124 mg/1 and 2.67 mg/1 respect--ively. This aquifer, thus, yields moderately potable waters, some needing defluoridetion.

High value of fluoride in well waters suggested that the possible source of fluoride may be the mineral fluorite and apatite found in granite. Waters occuring in granitic formation are invariably soft due to leaching of alkalis.

#### RHYOLITE:

Rhyolite occurs as hydrogeological formation around village Vandar, Lakhawas and Chatwara in south, south-west areas.

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Ground water is usually fresh i.e. bicarbonate to mix type and occurs under water table condition in weathered and fractured portion of these rocks.

#### TYPE OF WATER:

On the basis of concentration of three most mobile anions, such as chloride, bicarbonate and sulphate, the ground water of phreatic aquifer is classified into different types by projecting their percentage values on Piper's diagram (Fig.6). The following types of phreatic waters emerge.

				No. of	samples
1.	Bicarbonate	type	HCO3	50%	26
2.	Mix type	type	None	50%	28
3.	Chloride	type	Cl ·	50%	111
4.	Sulphate	type	so <sub>4</sub>	50%	0
			Total		165

The plate IV shows the areal distributions of types of phreatic water. It beers resemblance to isocone map plate-II. Their general chemical characteristics are given in table No.II and their associations with cations are shown in table No.III.

#### BICARBONATE TYPE:

It is fresh water, occupying area 1,2 & 5B of Piper's diagram and generally occuring where rein--fall and infiltration are relatively high. The major characteristics are low dissolved solids and low content of chloride. Only 15.7% water samples have shown bicarbonate type of character with either sodium or Ca + Mg as dominate cation. It occurs along hilly terrain of Jaswantpura. some area south of Raniwara, a small portion along river Sagi in Bhinmal, northern part of Saila where some recharge is contributed by Siwana hills, and in Jalore where river Jawai is contributing fresh water along the river course to a limited extent around Leta and Keshwana section. Limited extention of bicarbonate type of water is also observed along the course of river Fondi.

Water occuring near the foot hill zones and along river banks has mostly Ca -  $HCO_3$  or Ca + Mg -  $HCO_3$ type of chemical character with an average T.D.S. value around 500 mg/1 while that of Na -  $HCO_3$  type occuring adjoining to Ca+Mg-HCO<sub>3</sub> type has an average T.D.S. value around 800 mg/1. It is clear from the above that Ca+Mg-HCO<sub>3</sub> type water is comparatively fresh and close to the recharge area while that of Na+MCO<sub>3</sub> type is relatively more mineralised and away from recharge zone.

#### MIX TYPE OF WATER:

17.0 % well waters (Table-III) show the mix type of chemical character i.e. none of the anions has shown the concentration more than 50%. They mostly occupy the area 5 A and part of the area 3 close to 5 B of Piper's diagram showing intermediate chemical characters. It is further observed from plate IV that mix type of waters being transitional in character, occupy the area between bicarbonate type and chloride type particularly in Jaswantpure, Raniwara and Saila. In the eastern part, the well waters are in contineous contact with fresh water from catchment area. Whereas in the west these waters are gradually getting contami--nated by saline waters with the increase in sodium and chloride. In the Charau - Dadhal section the saline waters receive an influence of fresh water from Siwana area resulting in the formation of mix type of water along the northern bank of Jawai - Sukri river. In the central portion around villages Dhansa, Modran and Nun mix type of water is found, due to the recharge from river Bandi, however, in limited area.

It is also observed from the table-II that half of mix type of waters have bicarbonate more than either chloride or sulphate with T.D.S. value below 1,000 mg/1 and they are found mostly adjoining to Na - HCO<sub>3</sub> type of water. In the remaining, it is the chloride which takes the place of bicarbonate and their total dissolved concentration is more than 1,000 mg/1 indicating their saline nature occuring in close vicinity of chloride type of water.

This clearly shows that mix type of water is gradually changing its chemical character from  $HCO_3$ - Cl type to Cl -  $HCO_3$  type with the increase in total dissolved solids. Further the total dissolved solids of the mix type of water (Table-II) is ranging between those of  $HCO_3$  - type and Cl- type of waters.

- 19 - <u>TABIE - II</u> CAL CONTENTS IN DIFFERENT TYFES OF WATER OF FHEETIC AQUIFER	type 26 anelyses Mix type -28 anelyses Chloride type-111 anelyses Max. Average. Min. Nax. Average. Min. Mex. Average.	1090 635 535 3840 1435 1152 17280 4161 Abore Fatehpur Keonle Modren Baori	187 104 63 998 320 364 9123 1780 Chandur Pel Keonla Barethe Ecori	0 19 25 836 183 51 3547 418 Chore Petehpur Keonla Bokra Beori	579 227 79 1906 425 159 4380 894 Peoli Harmu Keonle Sakrie Dhenpure kidheni	80 21 Nil 600 146 Nil 600 119 Vender Dodiyeli Murtare Dayalpure Nimbla Bhinmel Ratunja etc.	8.00 2.07 Nil 9.6 2.43 Nil 14.20 2.45 Rejenwadi Beutre Hermu etc.
ENTS IN DIFFERENT 1	26 anelyses Mix Average. Min.		104			21	2.07
A CONTRACTOR CONTRACTOR CONTRACTOR	Bicarbone Min.	270 1090 Fanseri Ahore	17 Fenseri	e nil 90 Rempura <i>L</i> hore Penseri	132 379 Paharpur Paoli	Mil 80 Lakhewes Vander Keswana etc.	0.64 8.00 Rampure Rajenv Surewa Panseri
	Constitu- -ents in mg/1	SQT	Chloride	Sulphete	T	20N	وعقا

	<b>ทิ</b> ตtอไ	4	136 (82.5)	2( <b>%</b> ,2)	24(14.5)	3(1,8)	165 (100.0)
	ε <u>.</u>	Sulphate So4 50%					
FER		Su SO	) niř	lin	Lin	nil	lia (
O - EATIC AQUI		Chloride Cl 50%	103(62.5)	Ţ	6(3.6)	2(1.2)	111 (67.3) nil
WATER OF PHRE.	ominent anion	- Mix anions 50 %	17 (10.3)	e e lia	10 (0.1)	1(0.6)	28 (17.0)
THREE THREE TIL	,ŭ	Bicerbonete + Carbonate HCO <sub>3</sub> -+CO <sub>3</sub> -50%	13 (79)	2 (1.2)	11 (6.6)	ı	26(15.7)
CLASSACION OF CAPICNS AND ANIONS IN GROUND WATER OF PHREATIC AQUITER	S.Ku. Dominent cation.		l. Sodium + Potassium Ne <sup>+</sup> + K <sup>+</sup>	2. Celcium Ce <sup>++</sup>	3. Ge <sup>++</sup> + <sub>ME</sub> .++	4. Mix cetions.	

Ground Water Resources of Jalore District Part-II Hydrochemistry

#### CHLORIDE TYPE OF WATER:

× - .

The chloride type of waters, (67.3 % of the total analyses) are those in which chloride is dominant having more than 50 % of the total anionic value. They are characterised by high total dissolved solids, mostly above 2,000 mg/1 with sodium as dominant cation. This is the ultimate stage of cationic and anionic exchange reactions, generally existing in discharge zone.

They fall in central and extreme end of the area 3 of piper's diagram where mostly analyses of saline waters and sea water plot. Chloride waters are widely distributed, as seen from plate "', covering central part, western part including Runn, and major area of Ahore block where T.D.S exceeds 5,000 mg/1. <sup>H</sup>ighest salinity (EC 10,000 microsiemens/ cm ) in ground water is recorded around Runn area in north and north \_ west of Sanchore where ground water exhibits almost sea water · type of chemical character i.e. Na \_Mg \_Ca \_bo4 - "Co3 type. This area forms the discharge zone and practically there exists high water level conditions. This chloride type of water, particularly in south and north \_east of sanchore has high contents of potassium and nitrate indicating agricultural activitias in these areas. Salinity in ground water occuring around Rama \_ Bijli \_Ghana \_ Raithal section can be attributed to the slow rground water movement restricted by the Siwana hills ( on eastern side ) . However around village Rama few fresh water pockets, within this saline zone, are located, which appears to be due to some local The vicinity with the second 

Ground Water Resources of Jalore District Part-II Hydrochemistry

recharge condition in the vicinity of hills. Saline condition is also observed in south and south \_\_ eastern portion of . Jalore town. The area lies under shadow zone of Jalore hills where ground water movement is sluggish.

#### CATTONIC RATTO

The cationic ratios viz Na/Ca, Na/Ca+Mgand Ca/Mg in respect of types of water have been worked out and shown in tabular form below:\_-

Туре	Na/Ca Na/Ca+Mg			<u></u>		Ca/Mg					
of water.		<sup>M</sup> ax.	Æv.	Min.	<sup>m</sup> ax.	Av.	Mi	n.	Max.	<b>₽v</b> .	
Bicar-	0.22	9.05	2.79	0.1	5 5.0	02	2.12	0.4	1 2.	63	1.60
bonate									~ ~	~ ~	4 4 77

<sup>™</sup>ix 0.47 55.14 5.49 0.30 14.31 2.17 0.20 3.00 1.13 Chlo-0.97 131.73 10.92 0.58 16.83 3.61 0.02 3.15 0.95 ride.

On thorough scrutiny of the above data the following observations are made: 1. The minimum, maximum and average value of Na/Ca and Na/Ca+<sup>10</sup>g ratios show the increasing trend from bicarbonate type to chloride type of water. It clearly indicates that the increase in salinity of ground water is mostly contributed by sodium chloride being highly soluble and the product of arid environment. 2. The Ca/<sup>10</sup>g ratio shows the decreasing trend in minimum and average value from bicarconate

to chloride type while its maximum value shows the

rise. This further indicate that magnesium shows increasing trend with the increase in salinity while the calcium fluctuates within a limited range because of the low solubilities of its salts ( calcium carbonate and calcium sulphate ).

<sup>1</sup>n fact the trend in maximum value should **\*\*\*** have been similar i.e. in decreasing order but two analyses in mix type and one analysis in chloride type have shown the ratio above 2. This exception is due to the association of calcium with either chloride or nitrate. Both these salts are highly soluble.

It has been clearly brought out from the above discussion that the trend ( increasing or decreasing ) in the cationic ratio has significantly reflected on the chemical type of water.

### \_\_\_\_\_24 \_\_\_\_<u>CHAPTER - 3</u>

### QUALITY OF GROUND WATER IN DEEPER AQUIFER

Ground water in phreatic equifer in Sanchore, Bhinmal, west of Saila, north of Jalore and Ahore is saline. Naturally the deeper source has to be explored.

The chemical characteristics and quality of ground water of deeper aquifers have been studied from 49 analyses representing exploratory and obser-\_vation wells. Analytical results are given in appendix \_ II.

The plate IV is the modification of the comparative E.C. values of water samples from both deeper and phreatic aquifer. It depicts the over all salinity of ground water in the district.

As seen from the plates IV and 1, the quality of ground water from the deeper aquifer is less mineralised than that of phreatic aquifer in north west of Bhinmal around Jodhawas and punase, west of Saila around Alware and Dahiwa, south of Sanchore and near pladar, while it is comparatively more saline in west, north\_west and north of Sanchore, north of Ahore and Jalore and north-west of Raniwara near Chatwara and Digaon. However, very little difference is noticed in chemical quality of ground water of both the aquifers in central, Southern and eastern parts of Raniwara, "aswantpura and southern part of Jalore. Based on chemical analyses, the additional potential zone found around Jodhawas and punesa, and near The table IV indicates the ranges of salinity in ground water of deep aquifers encountered in various P.S. of Jalore district.

TABLE-IV TABLE SHOWING THE RANGES OF E.C. (in US/Cm ) FOR DEEPER								
			AQUIFE		n an an the second second	T Mit (1) - Se dan landsan yang selamatan dan		
P.S.	No.of analyses	0- 2000	2001- 4000	4001- 6000	6001- 8000	Above.		
Ahore	10	1	4	·····	2	3		
Bhinmal	10	3	1	2	1	3		
Jeswantpi	ara 6	5		-	1	-		
Jalore.	4	2	2	-	-	-		
Raniwara	. 5	3	-	-	-	2		
Saila	7	4	2	-	-	1		
Sanchore	7		-	2	2	3		
Total	49	18	9	4	6	12		
%	<b>%</b> 100	36.7	18.4	8.2	12.2	24.5		

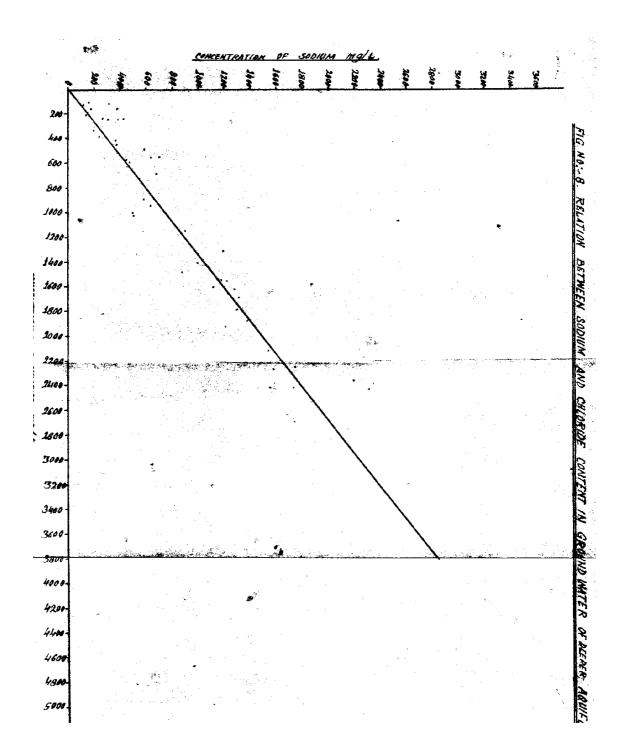
On perusal of above table it is seen that 55% c? the exploratory and observation wells have yielded fresh to tolerable quality of water while 26% wells have highly saline waters which can not be used for any purposes and mostly found occuring in Ahore, Bhinmal and Sanchore, where phreatic aquifer is also saline.

The salinity is mostly contributed by the sodium salts but in highly saline waters (E.C. 10,000 US/cm) substantial amount of calcium and magnesium salts are observed. Nitrate content except in few analyses is less than 20 mg/1 and unlike phreatic aquifer it does not show any relationship with either Salinity or hardness or potassium.

Fluoride is compara-tively lower in ground water of deeper aquifer. Twenty two (22) analyses have fluoride content more than 2mg/1. Low Salinity ground water of Jaswantpura, Bhinmal and Jalore has fluoride content more than 2 mg/1. The deep ground water at Rejanwadi, Jalore has 7 mg/1of fluoride. Where in phreatic aquifer the fluoride content is 8 mg/1. Deeper saline water of Ahome has also high contents of fluoride (Raithal -F.10.0 mg/1, Tarwara F-7.7 mg/1 and Sankarana F-4.28 mg/1).

Ground water is hard except in Saila and Raniwara where hardness of ground water is below 200 mg /1 as cCaCo<sub>2</sub>.

As seen from the graph, figure 7, cf E.C verses chloride, chloride content is found increasing with the increase in salinity up to 8,000 us/cm but with higher salinity value i.e. above 8,000 us/cm the proportional increase in chloride is less. Sulphate is gradually found increasing in high salinity vaters, Further it is observed from the graph, Fig.E., of sodium verses chloride, which looks like a contier for higher values, that the source of salinity is not simply the addition of sodium chloride but further modification by base exchange between formation materials and circulating water. This is evident from the high contents of calcium and magnesium in these high salinity waters.



The ratio of total dissolved solids to electrical conductivity varies between 0.55 to 0.93 with an average value of 0.625. However, for saline waters (EC 10,000 us/cm), the average value value is 0.69. But in high salinity waters (EC

30,000 Ms/cm) of Dugnri, Raithal and Tarwara the ratios are 0.8, 0.83 and 0.93 respectively. However the formula TDs (mg/1) = 0.61 x E.C at  $25^{\circ}$ C. can be used for rapid calculation of deeper ground water having EC value less than 10,000 us/cm.

# CHEMICAL CHARACTERISTICS IN RELATION TO GEOLOGICAL FORMATION:

The influence of geological formation on chemical characteristics of ground water is profound. The chemical changes brought about by different water bearing formations in the ground water of phreatic aquifer have already been discussed in the earlier chaper.

Three water bea-ring formations namely younger alluvium, older alluvium and granite were further explored for deeper aquifer so that some confined and concealed fresh water source may be located. The search brought to light the presence of tertiary (?) as a promising water bearing formation around Jodhawas, north-west of Bhinmal, which has yielded free flowing deep ground water of tolerable quality. However it has yielded saline water further down stream which can be used for livestocks.

					- M M	TORNAL TION
RANOP OF BURE URBRADEL S. Water beering No.of	t	Per-	Per-	Minimum -	mex1mum.	
No.formetion.	enelyses of eech aquifer.	861 867 •	Electrical conductivity in microsiemens per centimeter at 25°C.	Chloride Cl mg/l	#verence NO mg/3.	Fluoride F mg/l
l. Younger alluvium.	6	18.4	1080 - 50,000 Modren Dungri mein 8,725	154 - 23180 Modran Dungri 3,325	0 - 85 1.0 Saila Sankarna Bhains- 43 3.1	1 In
2. Older alluvium.	22 4	9.44	360 - 43,000 Jait- Raithal Pura, 7,188	18 - 16,628 0 Jeit- Reithel Umed- pure 2,358 22	0 - 105 Umed- Kuaber Pura 23	0 - 10.00 Rani- Reithel Ware 2.26
3. Granite	с В	16.3	450 - 37,000 Fanse- Chet- ri wera obs.	12 - 12,162 Fense- Chat- -ri. were 2,311	5 - 410 Bhin- Paoli mal. 76	1.08 - 7.60 Khan- Korita pur. 2.85
4. Tertiary + alluvium.	10	20.4	1600 - 14,700 Alwara Chajj- 7,515	270 - 5,140 Alwa- Chajjla -ra. 2,236	15 - 170 Batere Meda 63	0.80 - 2.0 Alwa- Chejjla ra. 1.31

- 29 -

Unlike phreatic aquifer there is a wide variation in chemical characteristics of deeper ground water of all water bearing formations. The minimum, maximum and average values of E.C., chloride, nitrate and fluoride are summerised in the table -V which on persual indicates the overall saline nature of the deeper waters.

The average E.C. and chloride content of ground waters exceed above 7,000 us/cm and 2,200 mg/1 respectively for all the formations. Their nitrate and fluoride contents are comparatively less than those of phreatic water. The maximum fluoride content of ground water from Tertiary is 2 mg/1 while it is above 5mg/1 in waters of the other formations. It is very significant.

The table VI gives the ranges of salinity in deep ground water of different water bearing formations.

RANGES OF	FE.C. (	US/CM)	IN DEEF	GROU	ND WATER	FORM	IATIONWI	<u>SE</u>
Water	No. of		R	ANGE	OF E.C.			
bearing	anely-	0-2000	2001-4000	-400	1-6000- <b>6</b>	- 000	8 0003	
formation	1.ses.				<u>ຣ 0008</u>	bove		-
Younger alluvium.		4	1	1	2	٩.	1	
Older al	luvium	9	7	-	1		5	
Granite.	8	4	-	1	1		2	
Tertiary	(+01-							
<u>der ellu</u>	vium10	1	1	2			4	
TOTAL:	49	18	9	4	6		12	
\$	100	36.7	18.4	8.2	12.2		24.5	

TABIE -VI

The formation wise breekup of salinity in deep ground water indicates that the ground water from older alluvium and tertiary are more saline than those of younger alluvium and granitic formations, where water circulates quicker in coarser formation material and in joints-and fractures respectively, thereby reducing the contact time.

The study of ionic ratios gives useful informations on chemical characteristics. The average Na/Ca ratio (10.5 to 12.0) does not show any signi--ficant variation for younger alluvium, granite and tertiary but it is comparatively less for older alluvium (8.7). Similarly Na/Ca+mg ratio is less for older alluvium, but its average So<sub>4</sub>/Cl ratio is higher (0.27) than those of other formations (granite 0.19, younger alluvium 0.16 and tertiary 0.18). It clearly suggests the calcareous nature of the formation materials in older alluvium.

The average Ca/Mg ratio is more or less same for waters tapping granite (1.14) older alluvium (1.18) and tertiary (1.15) while it is higher for younger alluvium (1.37).

#### TYPE OF WATER:

The analyses of deeper ground water like those of phreatic, are classified into different types by projecting their percentage values on piper's diagram (Fig.9) and their formationwise associations with cations are shown in table No. VII. Their chemical characterics are shown in table No. VIII.

	No. of samples
Bicarbonate typ	e. 8
Mix type:	8
Chloride type.	33

- 31 -

On perusal of piper's diagram (Fig.9) it is observed that bicerbonate type of ground waten from Jetpura (no.20.23; Raniwara) and Panseri (no. 33,34; Jaswantpura) are of Ca+Mg type with T.D.S less than 300mg/1 and are occupying the area 1 of the diamand shaped diagram. Such low salinity fresh water occurs where rainfall is heaviest. and water circulation within the zone of saturation is quicker. Both these tubewells are within in the catchment zone where geological formation has no control on chemical characteristics of water. The water moves out before any significant chemical change is brought about by the formation materials. The waters of Modran (no.31, Bhinmal ) and Raniwara (No. 43) are occupying the area 5 B. They are comparatively saline as the chemical characteristics change from  $Ce+^{H}E-^{H}CO_{3}$  type to  $Ne-^{H}CO_{3}$  type with the increase in TDS value around 700 mg/1. They are away from the catchment zone. However the Na-HCO3 type of water from Rajanwadi ( no.38-39-Jelore) is typical of <sup>J</sup>alore granite having TDS value of 1,200 mg/1 and fluoride content of 7 mg/1. The higher mineralisation can be attributed to semi arid to arid condition whereeven recharging water itself is saline . The source of fluoride is granite.

The mix type deep waters, from Alwara (NO.1), Bautra (No.6), Dahiwa (No.15, 1st zone), Kolar (No.25), Funasa (No.37), Saila (No.44) etc, mostly occupy the lower part of the area 3 close to 5 B of piper's diagram having transitional chemical characteristics with TDS ranging between 550 mg/1 to about 1,200 mg/1. They gradually change their chemical character from  $HCO_3$ - C1 to C1 -  $HCO_3$  type with the increase in mineralisation.

<ul> <li>cation.</li> <li>l. Younger alluvium. Na +K Ca+Mg Mix</li> <li>2. Older alluvium Na +K Ca+Mg Mix</li> <li>3. Granite Na + K Ca + Mg</li> <li>4. Tertiary. Na + K</li> </ul>	nt	Dominant	enion	Totel
	. Bicerbonate 50%	Mix none 50%	ix Chloride one 50% 50%	
5 6 8 8 5 6 7 8 8 1 8 8	I I	~ 1 1	١١و٧	יויס שו
6 E E E E E E E E E E E E E E E E E E E	₹ <b>~</b> ~ 1	KU 1 1	13 - 1	19 2 1 = 22
	K Mg	p-4 8 p-4	4 - 1	00 11 00 00
Ca + Mg Mix.	Me Me	∾ 1 1	ωιι	10 - - = 10
TOTAL	œ	σ	32	49

. .

	R MIXED TYFE Min. Mex. 550 2020 Bautra Bhinmel Golia 107 673 Peoli Dahiwa	R/NG.S. OF SOME CHEMICAL CONTENT IN TYPES CHEMICAL CONTENTN. TYPES CHEMICAL CONTENTN. TYPES CHEMICAL CONTENTInc.BICATSON/TE TYPECHEORILLE TYPEAverageBICATSON/TE TYPECHEORILLE TYPEInc.Min.Mex.Average.AverageMin.Mex.Average.Inc.2201190575574JeitpureRajanwedi57557440500JeitpureRajanwediBeutreDungri299122269014323180PenseriRejenwadi9014323180Obs.Obs.0bs.3624	E CHEMICAL CONTENT BICATEONATT TYPE Min. Nex. 220 1190 Jeitpura Rajanw Jeitpura Rajanw 0bs. 226 Panseri Rejenw	CONTENT IN NEX. LVE NEX. LVE Rejanwedi Obs. 226 Rejenwadi Obs.	IN TNFES CF Average. di 575 di 90 di	Min. CHIC Min. 574 Beutra Beutra Beutra	ER OF DEEFWR CHLOUILE TYLE Max. A0500 tre Dungri tre Dungri tre Dungri	.cuifer Averere. 7510 3624
223 Dehiwa	96	10	2 Fenseri	84 Reniwere 48 M	32	22 Beutra	6920 Tarwara	885
410 Faoli	100		Tr. Rajenwadi Obs.	25 Penseri		nil 170 Bagra,Daspa,M Modran,Rei- 0 -thal,Sanka- -rne,Umedpura etc.	nil 170 Bagra,Daspa,Meda Modren,Re1-0bs. -thal,Sanka- -rna,Umedpura etc.	34
7.00 2.49 Paoli	2.49		nil Jeitpura Penseri	7.00 Rejanwadi	2.60	αt	10.00 a Raithal etc.	2.22
446 189 Kolar	189		70 Rajenwadi	221 Raniwara 48 M	145	190 Pledar	5975 Heriyeli	1460

Chloride type of waters representing mostly older alluvium and tertiary occupy the extreme end of the area 3 of piper's diagram. They are saline waters and generally away from the recharge zone. The water of chatwara, Chajjala, Dungari, Raithal, Bharajun and Tarwara almost fall near the plot of sea water, exhibiting Na-Mg-Ca-ClSo<sub>4</sub> - HCo<sub>3</sub> type of chemical character.

The sole association of sodium as dominating ion in respect ofmix and chloride type of deeper water indicates the slow circulation resulting in base exchange, and concentration due to very poor permeability of formation (older alluvium, & tertiary).

While superimposing these three types of deep ground water on the map (plate-lv) showing areal distribution of chemical type of phreatic water it is observed that they are occupying the same zones which have been demarceted for phreatic water. However in sails the extention of mix type of character in deeper aquifer is observed upto Alware and Dahiwa where /pbreaticwater is chloride dominating.

Like phreatic aquifer the cationic ratios viz.Na/Ca, Ma/Ca+Mg and Ca/Mg have provided valuable information on different types of water. Their values are shown below in tabular form

Type of RATIO water. Na/Ca Na/Ce+Mg Ca/Mg Min. Max. 400. 34 20.17 6.18 Max. Ave Min Min. Bicarbo-0.34 0.22 11.96 3.63 051 nate Mix type 1.10 Chloride 1.50 16.61 10.43 0.71 32.16 11.12 0.76 9.98 5.75 0.503.22 1.5v 13.68 3.90 0.18 4.88 6.4 type

The following observations are made from the above table: 1. The minimum and the average values of  $N_2/2$ 

1. The minimum and the average values of Na/Ca and Na/Ca +Mg ratio show increasing trend from bicarbonate to chloride type of water. However the maximum value for mix type shows a decline instead of proportional rise, which can be attributed to calcareour nature of the formation material where mix type of water circulates.

2. Like phreatic aquifer, the Ca/Mg ratio shows the similar trend, the minimum and the average value show the declipe but the maximum shows rise.

The decreasing trend in minimum and average value from bicarbonate type of chloride type with rise in salinity indicates that calcium fluctuates within limited range due to low solubilities of its salts while like sodium, magnesium salts are soluble, so its concentration is likely to mcrease in chloride type of saline water.

Only one analyse of chloride type and two of mix type have Ca/Mg ratio above 2 otherwise the trend in maximum value should have been in expected decreasing order.

The trend (increasing or decreasing) in cationic ratio has provided useful informations on chemical type of water.

# CHAPTER - 4

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## SUITABILITY FOR DRINKING

Water used for domestic and drinking purposes should be free from any bacterial contents, turbidity, colour and odour. In Jalore district, ground water is assuming a great importance for domestic utilisation because rainfell is below . normal and surface storages are insufficient to meet the perennial requirements. Ground water (protected) generally do not contain suspended matter and bacteria, its use for domestic purpose has been solely considered on the basis of the chemical composition of its solutes or dissolved solids.

The concentration of emeny chemical substances in potable water varies widely from place to place so the rigidity of standards of quality can not be enforced. However the potability of ground water of the district has been adjudged mainly on the basis of drinking water standards recommended by the Indian Council of Medical Research (1975). A comparison of standards from World Health Organisation (International) along with I.C.M.R. has been given in table - IX showing maximum concentrations generally acceptable to consumers and also maximum allowable concentration beyond which potability is seriously impaired. The important cons--tituents like - total dissolved solids, chlorides, sulphates, nitrates, fluorides and hardness in ground water are discussed here.

#### TOTAL DISSOLVED SOLIDS:

The I.C.M.R. recommended an acceptable limit of 500 mg/l and maximum permissible limit of 1500 mg/l of T.D.S It has further relaxed the limit upto 3000 mg/l where better quality of water are not available, provided they do not contain harmful constitutents. It has been observed that people in arid and semi-arid regions drink waters with T.D.S. between 2,500 to 4,000 mg/l and no harmful physiolo--gical effects of permanent nature are noviced.

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Plate II showing the distribution of total dissolved solids in ground water depicts complete picture of salinity in phreatic and deeper aquifers. It reveals that fresh waters with T.D.S. less than 1,000 mg/l occur in Jaswantpura, Raniwara, and along the banks of river Sukri, while potable waters, having T.D.S. upto 3,000 mg/l are observed in parts of Saila, Jalore, Thore and Bhinmal. Non-potable saline waters (TDS 3000 mg/l) occur in Sanchore, north-west of Bhinmal west of Saile and north of Ahore, creating serious drinking water problem.

The table X gives the ranges of total dissolved solids in phreatic and deeper ground waters of different blocks of the district. It is observed from the table that 29% of well waters and 43% of tube well waters have TDS value less than 1,500 mg/l, the maximum permissible limit of I.C.M.R. Further 26% and 12% of water from phreatic and deeper aquifers respectively have been found to range between 1,500 to 3,000 mg/l.

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43% of the well waters, having more than 3,000 mg/l of T.D.S., are Widely distributed throughout the district except Raniwara and Jaswantpura, They should not be used for drinking. Similarly saline deeper waters (45%) from Bhinmal, Jalore and Sanchore should also be rejected. However these saline waters can be used for livestocks.

		- 39	E-I	<u>TABLE - IX</u>		
NEW AND OF CLARK (1975) STAFINEDS OF C	PHYSICAL D/ RDS OF	KND CHENIC/L	QUALITY OF	POPAEIEW.1	POTAEIEWATER COMPARED WITH	HILM O
No.Water clessification and substances.	Units W.	W.H.O. International Acceptable. 110wab	tionel( <u>4)</u>	Indi (ccentable.	Indien(B) ble Permissihl	, Travalal a
1. 2.	3.	4.	5.			-11
1. WITTER FOR GENERAL DOMESTIC USE	B					
1.1. Colour.	mgpt/L	5	8	ſ	25 25	\$
1.2. Turbidity	mhSio <sub>o</sub> /I	5	25	ت	25	
1.3. Taste		n.o.	n.o.	й.0.	n.o.	ł
1.4. Odour	ł	n.o.	n.o.	n.o.	n.o.	1
2. WATER OF SALINITY AND HARDNESS	Ω.					
2.1 pH	1	7.0-8.5	6.5-9.2	7.0-8.5	6.5-9.2	
2.2. Totel dissolved solids.	ug/L	200	1500	500	1500	3000(b)
2.3. Total hardness.	mgc $co_{\pi}/T$	n.m.	n.n.	300	600	, ,
2.4. Celcium, Ce	mg/I <sup>a</sup>	75	200	75	200	1
2.5. Megnesium, Mg	ng/L	50	150	50(e)	100	ł
2.6 Magnesium+sodium sulphete	mg/I	500	1000	3	ı	1
2.7. Sulphate, $SO_4$	mę/J	200	400	200	400	i
2.8. Chloride, Cl	ng/I	200	600	200	1000	ŀ
3.1. WITER APPECTING HUMAN HEALTH	H					
3.1.1. Fluoride, F.	mg/L	n.m.	1.5	1.0	2.0	ł
3.1.2. Nitrate, NO3	mg/I,	n.m.	30	20	сь.	100(c)

			- 94 -			
1. 2.	3.	4.	5.	6.	7.	8,
3.2. WLIER CLUSING TOXIC EFFECTS						
3.2.1. Iead,Pb	mg/L	n.m.	0.03	I	0.10	ı
3.2.2. Arsenic,4s	ng/I	n.m.	0.05	i	0.05	ı
3.2.3. Selenium Se	mg/L	n.n.	0.01	ł	L0.0	ı
3.2.4. Chromium (b+), Cr	mg/L	n.m.	0.05	ł	ł	8
3.2.5 Cyanide, CN	ng/L	л.ш.	0.20	ł	0.05	0.05
3.2.6. Cedmium, Cd	ng/L	n.m.	10.0	I	10'0	1
3.2.7. Berium, Ba	ug/L	n.m.	1.00	ł	,	ı
3.2.8. Mercury, Hg	mg/L	n.m.	n.m.	1	0,001	ı
3.2.9. Silver, 4g	ng/I	n.m.	n n	ł	ł	ł
3.3. WATER WITH NON-TOXIC METAIS						
3.3.1. Iron, Fe	ng/L	0.3	1.0	0.1	1.0	ı
3.3.2. Mengenose, Mn	m£/l	0.1	0.5	0.1	0.5	ł
3.3.3. Copper, Cu	mg/L	1.0	1.5	0.05	1.5	ı
3.3.4. Zinc, Zn	mg/L	5.0	15.0	t.	ı	ı
3.4. WITHE MITH ORGINIC POLLUTION	21					
3.4.1. BOD 5	me0 <sub>2</sub> /L	n.m.	9	â	ı	ł
3.4.2. FV (Oxygen obs.KMnO <sub>4</sub> )	mg02/L	н.	10	ŀ	Ł	1
o.4.3. £muonia, NH3	ng/l	n.m.	0.5	ı	ł	•
3.4.4. TOTAL NITROGEN, EXCLUSIVE NITRATE.	ng/L	ъ. в.	0.1	8		•

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1	

1. 2.	3.	4.	5.	6.	.7.	æ
3.5. WATER WITH ORGANIC POLLUTION INTRODUCED LETIFICIALITY.	ON					
3.5.1. Surfectants ABS	mg/L	0.5	1.0	• •	8	•
3.5.2. Organic matter as carbon in chloroform extract.	mg/L	0.2	0.5	1	1	ı
3.5.3. Phenolic substance as phenol.	n <i>e</i> /1	0*00T	0,002	0.001	0.002	1

NOTE: n.m. = Not mentioned.

n.o. = UnobjectionsDle.

- = Odour scele in U.S.K. 8
- 3
- = International standards for Drinking water, . W.H.O. Geneva, 1963.
  - = I.C.M.R. Standerds for Drinking Weter, India (1975) (E)
- = Not more than 50 mg/L, if there are 200 mg  $\mathrm{SO}_4/\mathrm{L}$ ; if there is less sulphate, magnesium upto 100 mg/L may be allowed. At the rate of 1 mg/L Mg for every 4 mg/L decrease in sulphate. (a)
  - = Total Dissolved solids relaxeble upto 3000 mg/L in cases where flternate sources are not available within reach. (q)
- = More information is required to prescribed a value but in no circumstances should the level exceed 100 mg  $NO_3/L$ . ં

	DISTRIBUTION	OF (T.D.S.	- JIT - JISS	I - X DISSCIMED S	• 4 • "YANIW - X DISTRIBUTION OF (T.D.S.) TOTAL DISSCRYD SOLIDS IN GROUND M. TER	NUD N. TLR
5. No. Block.	Source	No.of semules	0-500	Renge 500-1500	of T.D.S. in m <sub>E</sub> /1 1500-3000 / bove	1 me/1 1 bove 3000
l. Ahore.	Dug wells Tube wells.	25 10	, , , , , , , , , , , , , , , , , , ,	5.00	∞~	6
2. Bhinmel	Dug wells. Tube wells.	102	۱ اسم		ശപ	12 6
3. Jelore	Dug wells. Tube wells.	74 1	i	40	50	ю I
4. Jeswentpura	Dug wells. Tube wells	16	¢i	r-w	١٩	50
5. Raniwera	Dug wells. Tube wells.	27 5	na	5	41	50
6. Senchore.	Dug wells. Tube wells	37 6	~-i i	- <b>.</b>	10	25 6
7. Seile.	Dug wells. Tube wells.	24 7	1 اسم	ъъ	5	10 1
TO 17. L :	Dug Wells.	165	8 (4.85)	41 (24.85)	(26.67) (	72 (43.64)
	Tube wells.	49	4 (8.16)	17 (34.69)	6 (12.24) (	22 (44.7)

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MITRITE - NO3:

Nitrate poisoning has been confined to infants less than three months old but adults drinking the same water are not effected. Gastrointestinal troubles are reported when waters containing high nitrates (7 100 mg/l) are used for drinking.

Since the level of nitrate in drinking water is initial factor for determining its fatal effects on infants, the I.C.M.R. after due consideration has recommended 20 mg/l of  $NO_3$  in the water to be used for infants feedings. 100 mg/l is the limit for human consumption.

The distribution map of nitrate plate No.II for phreatic aquifer reveals that major part of the district having T.D.S. more than 3,000 mg/l contains more than 100 mg/l of NO<sub>3</sub> viz. north-east of Ahore, south-east of Jalore and major portion of Sanchore. In fresh water zone (TDS  $\angle$  1,000 mg/l) nitrate above 100 mg/l is observed towards north-east of Raniware. In general major part of the district except Sanchore F.S. have nitrate value in phreatic aquifer less than 100 mg/l. The range of nitrate as seen from the table XI indicates that nitrate in deeper equifer is comparatively very less than that of phreatic aquifer.

			- 44 -	2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2.		
	A	DISTRIBUTION	OF NITRATE	IN GROUND WITER	D W. TER	
No. Block.	Source	No.of			of NO <sub>3</sub>	in mg/l
		- Seuptes -	0-20	21 - 50	51 - 100	▶ 100
1. Lhore	Dug wells. Tube wells.	25 10	13 7	ЧQ	Ч 04	Б
2. Bhinmel	Dug wells. Tube wells.	22 10	ŝ	ЪЧ	1 60	C I
<b>3. Jalore</b>	Dug wells. Tube wells.	14 4	44	91	1	N I
4. Jaswent- pure.	Dug wells. Tube wells.	16 7	4 KU	500	nar d	50
5. Raniwara	Dug wells. Tube wells.	27 5	8 4	പെ	KU I	
6. Sanchore	Dug wells.	37 6	<i>ᲐᲐ</i>	KV 4	ш <i>т</i>	20
7. Saile	Dug wells. Tube wells.	24 7	БØ	80	4 (V	4 L
TOTL	Dug wells.	165	48 (29 <b>.0</b> 9)	43 (26.05)	18 (10.91)	56 (33.94)
	Tube wells.	49	28 (57.14)	8 (16.33)	8 (16.33)	5 (10.20)

#### FLUORIDE:

Fluorides are regarded as essential constituents of drinking waters for preventing dental carries in children. It has been observed that relative freedom from decay could be conferred by the presence of about 1.0 mg/l in drinking water but concentration over 1.5 mg/l cause mottled enamel. While regular intake of high concentration of fluoride may cause endemic cumulative fluorosis resulting damage to children and adults.

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Therefore, in the assessment of safety of a water supply with respect to the fluoride the total daily fluoride intake by the individual must be considered which inturn depends upon the climatic conditions. In view of the above observations the I.C.M.R. (1975) has suggested 1.0 mg/l as acceptable limit and maximum permissible limit of 2.0 mg/l in drinking water.

The distribution map of fluoride plate-III for phreatic aquifer reveals that concentration of fluoride above 2 mg/l in ground water having total dissolved solids less than 3,000 mg/l is observed around Bhinmal, south-west of Saila along river Sukari and major portion of Jalore. The table-XII gives different ranges of fluoride in well waters of various blocks of the district.

The area, having TDS value more than 3,000 mg/l in phreatic ground water with fluoride content more than 2-mg/l, are the upper portion of Ahore upto Shankhwali - Manpure, north-west and western

portion of Saila. In general it can be inferred from the distribution map that ground water in Jalore district has fluoride content more than 2mg/l. Jalore, Bhinmal and Ahore blocks are problematic area where several cases of Skeletal fluorosis have been noticed.

Fluoride in phreatic aquifer having TDS less than 3,000 mg/l, varies from 0 to 8 mg/l (Rajanwadi - Jalore) and in deeper aquifer it ranges from 0 to 10.00 mg/l (Raithal). It is observed from the table No.311 that fluoride in deeper aquifer is not as high as in phreatic aquifer.

The potable tubewell waters of Bhinmal -Golia (3.6 mg/l), Punasa (3 mg/l), Modran (5.32 mg/l), Korita (7), Rajanwadi (7.00 mg/l), Sankarna (4.28 mg/l), Paoli (7.0 mg/L) a Flader (2.00 mg/l) should be used only after defluoridation or by mixing water containing no fluoride so that fluoride content will remain below 2 mg/l.

- 42 -	DISTRIBUTION OF FLUORIDE IN GROUND WATER	Source No.of Renge of F in mg/l source samples. $\overline{0-1}$ $\overline{1-2}$ $\overline{2-4}$ > 4	ωm		<u>6</u> 00	Dug wells. 16 6 3 2 2 Tube wells. 7 3 - 2		01	Dug wells. $24$ 12 5 $\frac{4}{5}$ 2 Tube wells. 7 5 2 -	Totel Dug wells 165 $\begin{pmatrix} 60 \\ (36.36) \end{pmatrix} {}^{34}_{(20.61)} \begin{pmatrix} 42 \\ (25.45) \end{pmatrix} {}^{29}_{(17.57)} {}^{17.57)}_{(17.57)}$ Tube wells 49 ${}^{18}_{(36.75)} {}^{11}_{(22.45)} {}^{12}_{(24.49)} {}^{8}_{(16.33)}$
		S.No. Block So	1. Ahore Du	2. Bhinmel Du Tu	3. Jalore Du Tu	4. Jeswantpura Du Tu	5. Reniwera Du Tu	6. Sanch ore Du Tu	7. Seile Du Tu	Ă

## CHLORIDE:

All natural waters contain chlorides. The chlorides are usually taken in the form of common salt as a part of normal diet and are not considered harmful to the consumer unless suffering from dise--ases of heart or kidney. A salty taste is imported to water at chloride concentrations of about 300 mg/l and the tolerance to chloride of human being varies with climate.

Since there are no adverse physiological affect of chloride on consumers, the I.C.M.R. (1975) has suggested acceptable limit as 200 mg/l and maximum permissible limit as 1000 mg/l. The permissible limit appears to be safe for hot climate where it helps in alleviating salt depletion of body. Moreover in hot climatic area mostly saline waters are available for drinking and people are used to their salty taste.

It is observed from the distribution map (Plate No.III) of chloride in phreatic ground water that northern, southern part of Jalore, north-west and north-east of Ahore, western part of Saila, west and south of Bhinmal and around Sanchore block have ground water with chloride content more than 1,000 mg/l The chloride concentration increases as the selinity increases from south to north-west, north and west. The north-west and west appears to be a discharge zone. Most of the wells in Sanchore, Saila, Ahore, Jalore contain high chlorides which restrict their use for domestic utilization. The table gives the ranges of chlorides in the well waters of different blocks.

			- 49 -		
S.No. Block	Source	DISTRIBUTIO	DISTRIBUTION OF CHIOLIE IN GROUND WATER	E IN GROUND WITER	• ************************************
		semples.	0-200	200 - 1000	<i>k</i> bove 1000
l. Lhore	Dug wells. Tube wells.	25 10	M I	σr	13
2. Bhinmel	Dug wells. Tube wells.	22 10	<b>K</b> \ -1	2	12
3. Jelore	Dug wells. Tube wells.	14 4	КП	<b>K</b> 0 K0	- ∞,
4. Jaswentpura	Dug wells. Tu be wells	16 7	r0 4	. ~ 1	40
5. Raniwara	Dug wells. Tube wells.	27 5	17 33	91	40
6. Sanchore	Dug wells. Tube wells.	37 6	ا ہم	11	- 25 6
7. Seila	Dug wells Tube wells	24 7	К	12 3	) നല
Totel	Dug wells	165	35 (21.21)	55 (33 32)	75 (AE AE)
	Tube wells.	49	12 (24.49)	12 (24.49)	25 (51.02)

Further the analysis of table No. XIII depicts that the chloride content is high in deeper aquifer and about 51% well waters have chlorides above 1,000 mg/l and therefore are not suitable for drinking.

#### HARDNESS:

Hardness in water is usually judged, by the consumer by the amount of soap required to produce lather. Hardness is caused by the dissolved calcium and magnesium salts. Iron and aluminium also contribute towards hardness if present in large amounts. The temporary hardness includes that portion of calcium and magnesium that would combine with the bicarbonate and the small amount of carbonate present and can be virtually removed by boiling the water. Similarly the parmament hardness is caused by those amounts of calcium and magnesium that would normally combine with sulphate, chloride, and nitrate ions present in water. This portion of hardness can not be removed by boiling.

The hard water has no major harmful affects on the health of consumers, but the main deterimental effect of hardness is economical, hence the I.C.M.R. (1975) has promulagated the following limits.

> Highest acceptable limit - 300 mg  $CaCO_3/L$ Maximum permissible limit - 600 mg  $CaCO_3/L$ .

On reviewing the value of total hardness (appendix-I) of the phreatic aquifer, it is revealed

that 45% well waters have very high values (above 600 mg  $CaCO_3/L$ ) which are distributed through out the district.

Table No.XIV shows the distribution of hardness in phreatic as well as deeper aquifers of the district. It is observed from the table that hardness in water in deeper aquifer is comparatively less than phreatic well waters. The Dug well yielding water with hardness above, 1,000 mg CaCO3/L should not be used for drinking purpose. Thus well waters of Bala (TH-2, 255 mg  $GaCO_3/L$ ), Mede (1,615 mg Ca  $CO_3/L$ ), Dhanpure (4,500 mg CaCO<sub>3</sub>/L), Korka (1,386 mg CaCO<sub>3</sub>/L), Ankhol (2,677 mg CaCO<sub>3</sub>/L), Dungri (1,420 mg CaCO<sub>3</sub>/L), Amli (1,126 mg CaCO3/L), Bhawatra (2,032 mg CaCO3/L), Bishangarh (1,981 mg CaCO3/L) etc. are very hard. They are also seline (TDS > 5,000 mg/L), thus not potable. Water sample of well at Agdawa having hardness more than 1,000 mg/l but TDS less than 2,500 mg/l is to be rejected on the basis of high value of hardness and fluoride (F. 4. mg/L).

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	in . In	ů S	:		
	DISTRI	TABI DISTRIBUTION OF	TABLE - XIV		IN GROUND WITER
S.No. Bloek	Sources.	No.of semples.	0 - 300	Renge of TH i 300 - 600	$\frac{\operatorname{in}\operatorname{mg}/1}{0} > 600$
l. Ahore	Dug wells Tube wells.	25 10	50	36	14 2
2. Bhinmal	Dug wells. Tube wells.	22 10	CO 4	, r4 r4	പ ന ന
3. Jalore	Dug wells. Tube wells.	14 4	40	50 60	
4. Jaswantpura	Dug wells. Tube wells	16 7	ſΩĸ∩	80 K	<b>7</b> 7
5. Raniwera.	Dug wells Tube wells	27 5	14 3	Ч 0/	
6. Sanchore	Dug wells Tube wells	37 6	4 ખ	<b>о</b> н	24 4
7. Saila	Dug wells Tube wells	24 7	6	۱ ف	. 11 1
Total:	Dug wells.	165	47 (28.48)	43 (26.06)	75 (45)
	Tube wells.	49	21 (42.86)	11 (22.45)	17 (34.69)

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## SULPHATE :

Higher concentrations of sulphate (above 400 mg/l) in drinking water may have a laxative effect on new users but normally do not effect the teste. The I.C.M.R. (1975) has recommended an acceptable concentration of 200 mg/l and 400 mg/l as maximum permissible concentration.

The table No. XV shows the concentration of sulphate in water samples collected from different blocks of the district.

On going through the table it is seen that 24.24% of samples from phreatic aquifer and 32.65% of samples from deeper aquifer have sulphate content above 400 mg/l and as such they are not safe for drinking. The high sulphate content was observed mostly in high saline waters of Bhinmal and Sanchore blocks. Only in one analysis of Murtara Silli of Jaswantpura both sulphate and magnesium have exceeded the limits of potability. Nitrate is also high 600 mg/l, therefore it should not be used for drinking.

It is inferred from the above discussions that drinking water is a problem in major part of Sanchore block, north-west of Saila, north of Ahore, central part of Bhinmal, and southern part of Jalore as some of the chemical constituents like NO<sub>3</sub>, F, Cl, TDS etc. in ground water are beyond relaxable and harmful levels and unfit for human consumption. Water treatment processes to remove them are costly. So - 54 -

far no concentrated attempt has been made to study the physiological effects of these excessive salts on local inhibitants. However, dental and skeletal fluorosis have been reported in Jalore and Ahore blocks.

A source, free from harmful chemical constituents, within the area has to be found out and protected, else better quality of water has to be transported at a great economic loss.

		-11			
	DISTRIBUTIO	N OF SU	TEN CLE IN	DISTRIBUTION OF SULFY (TE, IN GROUND WITER	
S.No. Block	Source No	No.of		Renge of	SO4 in mg/l.
	S	semptes.	0-200	200 - 400	> 400
l. Lhore	Dug wells Tube wells	25 10	10	40	11 31
2. Bhinmal	Dug wells Tube wells	22 10	2-2	9 60	0.4
3. Jalore	Dug wells Tube wells	14 4	o 4	41	<b>н</b> I
4. Jaswantpura	Dug wells Tube wells	16 7	11 2	KU I	50
5. Reniware	Dug wells Tube wells	27 5	23 3	ᡨ᠇	1 –1
6. Senchor	Dug wells Tube wells	37 6	0T 1	15 1	12
7. Saila	Dug wells Tube wells	24 7	10 4	on∾	بے ر <i>ی</i> ا
Totel	Dug wells	165	80 (48.48)	45 (27.27)	40 (24.24)
	Tube wells	49	24 (48.98)	9 (18.37)	16 (32.65)

- **55 -**TABLE - XV

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#### CHAPTER-6

## SUITABILITY FOR IRRIGATION

It is well known fact that the prosperity of an area is directly related to its irrigated In semi-arid region like Jalore, the main lands. source of irrigation is ground water because rainfall is limited and no other source of irrigation is available. Development of soil salinity and alkalinity are intimately related to the chemical composition of irrigation waters which in turn effects the crop productivity. Therefore, the quality of ground water is very important in determining its suitability for irrigation. Besides the chemical quality of ground water, the following factors should also be given due considerations for determining the quality standards of irrigation waters. Factors effecting the quality of water for irrigation are:-

- L. <u>CLIMATE</u>: It has a bearing on selinity in soils due to precipitation and evaporation. Selt tolerance of crops also depends upon climatic conditions.
- B. <u>TOPOGRAPHY</u>: It is responsible for drainability of soluble salts in soils.
- C. <u>SOIL CHARACTERISTICS</u>: Texture, nature of clay minerals, organic matter, pature and concentration of salts present in soils are few important characteristics which determine the effect of irrigation water on the soil.

D. <u>SALT TOLERANCE OF CROPS</u>: Type of crops to be grown also determines the suitability of irrigation water. Crops like Bajara,Cotton, Jawar etc. can be grown with relatively saline waters on saline soils while the some water may not be suitable for irrigation of the salt sensitive crops (Citrous, lemon etc. ) even on good soils.

# E. EFFECT OF FERTILIZERS AND MANURES:

Though the use of fertilizers and manures increases the crop productivity, their effects on soil properties with respect to the type of fertilizers being applied to the soils can not be ignored.

# F. WATER AND SOIL MANAGEMENT PRACTICES:

These control the accumulation of soluble salts in soils depending upon the texture of soils, water applied and drainability of the area.

# CLASSIFICATION OF IRRIGATION WATER:

The water quality criteria for irrigation is a complex subject and as such it has never been possible to have a fool proof criteria because the growth of a specific crop depends on verious factors and not merely on the chemical composition of irrigation water. But it was soon realized that the factors enumerated above are responsible for this complexity. Thus the criteria which have been proposed by various workers for assessment of quality of irrigation water may not be applicable under particular condition. However, for classification of the ground water for irrigation purposes, chemical composition of the water alone has been taken into consideration and following parameters for evaluating the suitability of water for irrigation have been studied. - 1997年の第二日の1997年の第二日の1997年の第二日の1997年の第二日の1997年の第三日の1997年の第三日の1997年の第三日の1997年の第二日の1997年の第二日の1997年の第二日の1997年の第二日の 1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997年の1997

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1.	Total soluble salts evaluated as Electrical
	conductivity.
2.	The proportion of cations viz. SAR and
	sodium persent.
3.	The relation of bicarbonate to alkaline
	earths.

The final recommendations for irrigation use have been considered on the basis of water Rating criteria proposed by Anonymous (1972), as it takes into account all the related factors viz. soils, waters and crops.

age of age of amples c Deeper 36.7 36.7 36.7 18.4 18.4 10.2 10.2	OFPercentage ofPercentage ofWater samplesPhreatic Deepaquifer. aqui26.726.726.721.88.29.79.7	XYI XVI XVI XVI VICONG WITH DISTRIBUTION OF FET Remules For a guifer, equifer, equif	6 WITH DIST No.of wat Samples Fhreatic 44 40 40 36 16		<ul> <li>SALINITY IN GROUND WARKS LABLE</li> <li>SALINITY IN GROUND WARKS LABLE</li> <li>SALINITY IN GROUND WARKS LABLE</li> <li>Crops to be Upper lanssib limit of for set for set for set for init for set for set for for set for init for set for set for for for set for for for set for for for set for for set for for set for for set for for set for for for set for for for set for for for for for for for for for for</li></ul>	
26 R	א קר	۲. ۲	00	BOOD		soils that have excellent internal & sunface Amainana
10.2	9.7	ъ	16	6000 8000	Semi tolerant. Tolerant.	4.Light textured soils having clay content of less than 10% soils that have eventiant
8.2	21.8	4	36	4000 6000	Semi tolerent. Tolerant.	Medium textured soil had clay content of 10-20% that are very well drai internally & have good drainage system.
18.4	24.2	σ	40	2000	Semi tolerant. Tolerant.	Heavy textured soil having a clay content of 10-20%, soils that are well drained interna-lly & have a good surface drainage system.
36.7	26.7	18	44	1500 2000	Semi tolerant Tolerant.	• Deep black soils and alluvial soils having a clay content more than 30% soils that are fairly to moderately well drained.
7.	6.	5.	4.	3.	2.	<b>1</b> ••••••••••••••••••••••••••••••••••••
age of amples c Deeper . aquifer.		er De∉ ∾er aquifer		per per- ssible nit of C.of wete r safe use r irrigat	to be	of
	OF	DISTRIBUTION	HTIM 9			NAMES AND WATTER CUALITY RATINGS (B

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- If the soils have impeded internal drainage either on account of presence of the harf pans, unsuelly high emounts of clay or other morphologic reasons, table at no time of the year is with in 1.5 metres from proposed limits of water quality it is pasumed that the the water table does come up within the root zone the ebove limits need to be reduced to half the sbove values. 1. In the above ground water surface. If പ് പ NO.15
  - for advisory purposes, the limit of water quality should again be reduced to half.
    - If the waters contain soluble sodium percentage more than 70, gypsum should be added to soil occessionally.
      - 4. If supplemental canal irrigation is available waters of fligher electrical conductivity could be used profitebly in lean periods.
        - 5. The above scheme act only as guidelines and the actual user should keep in view the specific local conditions.

However, passing reference has been made to classify the analyses based on modified  $\mu$ .S. selinity Laboratory diagram. According to it, nearly 40% waters fall in C<sub>5</sub> -S<sub>4</sub> (EC 4000 - 6000  $\mu$ S/cm) and C<sub>6</sub>-S<sub>4</sub> (EC 6000 - 10000  $\mu$ S/cm) group and 10% in C<sub>7</sub>-S<sub>4</sub> (EC 10000  $\mu$ S/cm) group, hence can not be utilised on most of the soils for irrigation purpose. But rest of the 50% analyses are falling under  $\theta_2$  to C<sub>4</sub> class of EC and S<sub>1</sub> to S<sub>3</sub> and even S<sub>4</sub> class of sodium adsorption ratio. They can be utilised with due precautions on most soils, however on clayey soils management practices are to be followed.

Since the soils of Jalore district are mainly sandy, loany sand and sandy loam, the classification proposed by Anonymous (1972) holds good as a guideline under the conditions prevailing in the district.

## ANONYMOUS WATER RATING CRITERIA (1972):

It is based on field experience of various agricultural experts working on utilisation of saline water under various soil conditions and different crop patterns. It takes into account the salinity hazard in terms of electrical conductivity of water and alkalinity hazard as sodium percentage and residual sodium carbonate.

## SALINITY HAZARD - ELECTRICAL CONDUCTIVITY:

165 and 49 representative analyses of ground water from phreatic and deeper aquifer respectively are grouped in different E.C. ranges and are presented, percentage wise, in table XVI.

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Distribution of water samples of different EC ranges shows that 72.7 percent of analyses are equally distributed in EC range upto 6000 US/cm while 17.6 percent analyses show salinity more than 8000 US/cm and the remaining 9.7 percent have selinity between 6000 to 8000 US/cm.

Plate IV based on periodic water analyses from phreatic as well as from deeper aquifers, depicts that the ground water in the district is of moderate to high salinity. Distribution of EC ranges in water samples of different panchayat samities show that ground water in Jaswantpura, Raniwara, part of Saila and part of Jalore is comparatively of low salinity where as ground water in Bhinmal, Sanchore, Ahore is of high salinity. The entire western part of the district has saline ground water having EC value more than 8000 JJS/cm. However, localised fresh water pockets, usually perched water, observed here and there in this salinity infested western part but they are of no significance for any meaningful utilisation.

Soils are mostly sandy, loamy send and sendy loam in textures with depth generally exceed--ing 60 cm. Water table is fairly deep. Hence, use of waters upto 8000 MS salinity for irrigation is not likely to result in increase of salinity of the irrigated soils except in certain isolated patches of high clay content. Waters with EC more than 8000 JS/cm should not be generally used. However, in certain area high salinity waters are being used for growing salt tolerant crops mainly Kharchia wheat, by adopting proper irrigation practices. Salt tolerant crops viz. wheat, rayda, barley, sunflower etc. can be successfully cultivated with these waters with appropriate level of fertiliser application and providing 30%50 percent additional irrigation water for leaching. Normal to above normal rainfall occuring in the district can be considered sufficient for leaching soluble salts accumulated in the soils as a result of irrigation with saline waters.

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## SODIUM HAZARD:

The sodium hazard in irrigation waters was evaluated on the basis of soldum to other cationic ratios such as SAR, sodium percentage and Na/Ca or Na/Ca+Mg. Sodium in relation to other cations of the water, effects the soil characterics. Soils irrigated with high sodium water become impermeable and hard creating serious drainage problem. It also deprives the plants from available micronutrients for maintaining healthy condition. On the other hand a low ratio tends to maintain the soils in good condition favourable for good crop.

#### SODIUM/CALCIUM AND SODIUM/CALCIUM & MAGNESIUM VALUES:

If the ratios exceed 1 in irrigation waters, socium accumulation tends to become a problem in the soils. It is particularly true for the soils having high clay content but on sandy soil water with high ratios can be used. It is less than 1 in about 10 percent of analyses thus remaining waters, when used for irrigation can cause sodium problem.

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# SOIDUM PERCENTAGE - Nas

Anonymous water rating criteria (1972) has indicated that waters with sodium percentage more than 70 create sodium accumulation in soils. As seen from table XVIII more than 40 percent and 65 percent analyses for phreatic and deeper respectively have sodium percentage above 70 and their use for irrigation needs suitable amendment. In such cases application of gypsum directly to the soils is effective.

# RESIDUAL SODIUM CARBONATE - RSC:

Residual sodium carbonate is often found in appreciable quantities in waters of low salinity. When waters containing high RSC are used for irrigation, they effect the soil characteristics edversely because of their strong alkaline nature.

The water rating criteria has indicated that waters with RSC up to 2 meg/1 can be used for irrigation without needing any amendment, but when exceeds this value, grinded gypsum at the rate of 8.5 quintal per heatare meter of water is to be applied to counter match the effect of 1 meg/1 RSC of irrigation water.

It is seen from the table XVII that 63% and 61% analyses from phreatic and deeper aquifer respectively do not show any RSC value while 25% and 27% analyses respectively have RSC value above 2 meq/1 needing corrective measures like application of gypsum. The table further indicates that the RSC toxicity is high in low EC waters (< 4000 JS/cm) as compared to high salinity waters. Sodium hazard, contributed by any of the above discussed properties can be reduced to a minimum level by correct application of gypsum to the soils as suggested.

The discussions, on salinity and sodium hazards of the ground water of Jalore district, have lead to the conclusion that with proper soil-water plant management the ground water with E.C. less than 8,000 JJS/cm can be profitably used for irrigation purpose mainly for semi-tolerant and tolerant crops as the soil conditions are favourable.

- 65 <del>-</del>

							TABLE	IIN -	IIA					
	DIS	TRIBUT	ION OF	SOID	UM PEI	ICENT ANI	I SEN	TYNG	SOD.	IUM C	<b>IRBOI</b>	T N	NT	DISTRIBUTION OF SOIDUM PERCENT AND RESIDUAL SODIUM CARBONATE IN GROUND WATER
S. E.C. DS/cm No.	ſ	SODIU	SODIUM % (No. of analyses atic Desner	(No.	of ar	lelyses)	RSC (No. Threatio	SC (]	No. 01	RSC (No.of analyses) hreatin	.yse	<u> </u>		
	lo Q	¥02	Total	70% 700	70%	Total		5	1 1	Total	1 0 2	-	~	Total
<b>1.</b> 0 - 2000	38	9	44	11	~	18	12	16	16	44	ŝ	ŝ	œ	18
2. 2000 - 4000	25	15	40	, <b>N</b>	2	ი	26	ĸ	11	40	4		4	ნ
3. 4000 - 6000	19	17	36	1	4	4	29	0	7	36	ŝ	0	<b></b> 1	4
4. 6000 - 8000	7	σ	16	1	5	ŝ	11	4	4	16	5	-0-	0	5
5° 8000	6	Ŗ	29	4	<i>о</i> л	13	26	0	m	29	13	· ~ •	0	13
TOTAL:	86	67	165 .	17	32	49	104	20	4	165	30		12	49
8	9.4	40.6	(1)	34.7 65.3	65.3	٢	63.0 12.1 24.9	12.1	24.5		61.2 12.2 26.6		26,	9
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÷ 66 F

### CONCLUSIONS

- 67 -

Waters of Jalore district, from phreatic and deeper aquifer have been analysed chemically and studied for chemical properties of water bearing formations and chemical types. These have been further interpreted for domestic and irrigation utilisation.

Phreatic water occuring in three major water bearing formations viz. younger alluvium, older alluvium and granite is fresh to saline. Saline water, with E.C. above 8,000 US/cm occurs in Sanchore, north of Abore, south-west and northeast of Jalore and part of Bhinmal (Plate-I.).

Ground water occuring in younger alluvium covering an area around the confluence of Luni and Jawai - Sukri rivers is less mineralised having bicarbonate, bicarbonate- chloride to chloride type. Saline ground water occurs in discharge zone, west of Sanchore.

Older alluvium, distributed mainly in north-eastern, south western, western and central parts of Jalore mostly yields sodium chloride type saline to highly saline water. More than 30% of analyses from this formation have high contents of nitrate, fluoride and chloride, prohibiting their use for domestic utilisation.

Both fresh and saline waters occur in granites. Fresh to moderately saline waters occuring near the foot hill zone of Jalore, Jaswantpura and Raniwara are soft with high content of fluoride. Saline waters occuring around Bhinmal and Ahore are of chloride type. Defluoridation or dilution by low fluoride water is suggested for bringing down the harmful level of fluoride before the water is supplied for drinking.

The gradual change of chemical type with the increase in total concentration from Siwana hills to south-west (Bagora) and from foot hills zones of Jaswantpura and Raniwara to north-west and west is confirming to flow direction.

Deep ground waters are hard and more mineralised in Sanchore, Ahore and Jalore but around Jodhawas, Punasa in Ehinmal and near Alwara- Dahiwa in Saila they are comparatively less mineplised. The chemical contents, like TDS, Chloride, Sulphate etc. of deeper ground water representing older alluvium, younger alluvium and granites are comparatively higher than those of phreatic ground water. However the concentrations of nitrate and fluoride are compara--tively less.

The ground water from the tertiary, hither to concealed, but now a promising potential aquifer has a tolerable quality of water (Jodhawas, Bhinmal) and can be exploited for both agricultural and domestic utilisation.

The change in chemical type of deeper ground water i.e. from bicarbonate to chloride type is well in confirmation with that of phreatic water. However, the ground water of phreatic aquifer from Dahiwa and Alwara are chloride type but in deeper water they are either bicarbonate type or mixed type.

The suitability of ground water for drinking purposes was judged on the basis of standards recommended by I.C.M.R. (1975). Only 29% and 42% analyses from phreatic and deeper ground water respectively fall under permissible level of TDS i.e. 1500 mg/l though some of them are not potable because of their high contents of nitrate and fluoride. However, 43% and 44% analyses from phreatic and deeper ground water respectively are unsuitable. Their total dissolved contents have exceeded the value of 3000 mg/l. Majore part of Sanchore, northwest of Saila, north of Ahore, central part of Bhinmal and southern part of Jalore have such unsui--table waters unfit for human consumption. Drinking water is a major problem of these area.

The recommendations on suitability of ground water for irrigation, are based on anonymous water rating criteria (1972) considering both salinity and sodium hazards. According to the classification ground water from phreatic and deeper aquifer having EC value upto 8000 JJs/cm can be used for growing tolerant and semi tolerant crops by adopting proper soil water plant managements. 17.6% and 26.5% analyses from phreatic and deeper ground water respectively are not recommended as their E.C. value have exceeded above 8,000 JJS/cm.

\* \*\*\*

	RSC	19	8.86		1 <sub>e</sub> 24	· 1	2,30	, <b>.</b>
	Na%	18	83.1	78, 55	82,9	72.07	74.2	64.8
	<sup>2</sup> Htto <sup>-1</sup> No <sup>-1</sup> P TH Na% RSC <sup>3</sup> mg/ mg/	16 17	4 <b>.</b> 0 140	3 <b>.</b> 2 2255	5. 32330	4,4,4670	0.32 353	1 <b>•0</b> 361
щ		15		9 <sub>4</sub> 33 0 <sub>4</sub> 50 569 31	•		2 <b>.</b> 25 140 (	0.24 15 ]
APP ENDIX- I ER	HRD 1	14	11,77 0,16 681 10	9, 33 569	7.84 478	8,68 530	9,37 572	6 <b>.</b> 37 388
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M PHRE ER (UP)	ម	F	153 153	176 <b>.</b> 6( 626:	3.22 24.29 39 861	257.0 9123	2 <b>.</b> 75 14 <b>.</b> 18 33 <b>,</b> 503	13 <b>.</b> 23 469
ED FROI ICT PERLIT	1 2 + 6 W	10	1.56 1.25 31 51	9-80 35.30 176-60 19.50 196 429 6262 937	3 <b>.</b> 22 39	32 <b>,</b> 506 <b>6,</b> 90 257 <b>,0</b> 651 741 9123	2 <b>.</b> 75 33	2• 34 28
DLLECT DISTR	្ម ខ្ម ខ្ម ខ្ម ខ្ម	6	1 <b>.</b> 56 31	9,80 196	3 <b>. 38</b> 68		<b>4.</b> 32 86	0 <b>.</b> 10 · 4.89 4 100
SAMPLES COLLECTED OF JALORE DISTRICT ALL EQUIVALENT PERI		ω	3 0.07 3 3 3 3	0_64 25	0 <b>,</b> 15 6	0 <b>.</b> 13	0.48 19	0 <b>.</b> 10
LYEES OF WATER SAMPLES COLLECTED FROM PHREATIC AGUIN OF JALORE DISTRICT EXERESSED IN MILL EQUIVALENT PERLITER (UPPER VALUE) MILLIGRAMS FER LITER/(LOWER VALUE)	( + c) 1. X	1	1 <b>4.1</b> 325	7 <b>•75</b> 167•46	32.67 750	241.40 5550	21.74 500	13,48 310
OF WA SED II GRAMS	i Ha	9	7,85	7.75	1.7	7.5	7.8	7.7
	ECX10 <sup>6</sup> TDS by pH Na <sup><math>+</math></sup> K <sup><math>+</math></sup> Ca <sup><math>+2</math></sup> Mg <sup><math>+2</math></sup> (Micro-calcu- Siemens/lation.	2	1090	11730	2560	17280	1980	1440
CHEMICAL AR		.+	1710	18300	4000	27000	3100	2250
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	1.2		E L	2.	en.	4.	ີ້ ທີ	<b>•</b>

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	14	9.21 562	4 <b>.</b> 61 281	6.27 383	4 <b>,</b> 90 299	12 <b>.</b> 84 783	3, 53 215	4 <b>.</b> 31 263	12 <b>.</b> 94 790	10,68 652	19.21 1172
	: 8	i t	t	1			2	ı	ı	•	
	12	<b>17,61</b> 846	6 <b>.</b> 67 320	1,56 75	6 <b>.</b> 67 320	7 <b>.</b> 86 378	58 <b>.</b> 60 2819	0 <b>.</b> 73 35	28 <b>.</b> 10 1350	15 <b>.</b> 14 727	9 <b>.</b> 68 465
		55,66 55,66 1973	29,11 1032	21.68 767	31, 37 1112	29 <b>.</b> 01 1029	64 <b>.</b> 44 2290	4 <b>.</b> 73 168	74 <b>.</b> 60 2648	64 <b>.</b> 26 2280	23 <b>,</b> 25 824
	10	16.54 201	9.73 118	6 <b>.</b> 24 75	9 <b>.</b> 88 120	3 <b>,</b> 9 <b>3</b> 48	16 <b>.</b> 32 198	1•04 13	11.42 139	17,58 214	6 <b>.</b> 82 83
4	6	10.97	10,97 220	6.08 124	11.70 234	2 <b>.</b> 76 56	19 <b>.</b> 04 381	2, 76 56	12 <b>.</b> 50 250	13 <b>.</b> 26 266	3 <b>.</b> 74 76
		0.15 6	0 <b>.</b> 30 12	0 <b>.</b> 07 3	0.43	0 <b>.</b> 05 2	0 <b>.</b> 92 36	0 <b>.</b> 07 3	0 <b>.</b> 15 6	0 <b>.</b> 32 12	0 <b>.</b> 15 6
		55,45 1275	23 <b>.</b> 05 530	18,96 390	21 <b>.</b> 31 490	43 <b>.</b> 49 1000	86 <b>.</b> 99 (2000	5 <b>.</b> 65 130	92,42 2125	58 <b>,</b> 75 1351	40 <b>.</b> 23 9 <b>2</b> 5
		1.1	7.55	7.6	7.8	7,85	7,65	7.6	1.9	7.5	8,1
	5	5120	2655	2050	2880	3200	6720	650	7100	5570	3072
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13 19	24.55 -	7.32 639 83.6 0.74	92 <b>.62 5.19</b>	56,04 -	61 <b>.</b> 0 -	43 <b>.</b> 6 –	83 <b>-</b> 39 <b>-</b>	63 <b>.</b> 72 7 <b>.</b> 59	71.51 1.50
16 17	24,55 0,64 1906	7.32 639	5 <b>- 32</b> 265	. 873	L <b>-</b> 64 689	• <b>32</b> 525	1•32 766	<b>3.6</b> 4 135	1.64 278
15	7,25 450	0, 56 35	9 <b>.</b> 67 600	0.66 10	0.16 10	4.27 265	0.24 15	ŧ	0 <b>.</b> 40 25
14	11.56 705	13 <b>.</b> 52 825	16 10.49 640	5, 49 335	8 7.64 0.16 - 466 10 1	3 <b>.</b> 63 221	78 4 <b>.</b> 51 275	10 <b>.</b> 29 628	7 <b>.</b> 06 431
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11 11 11 11	28 <b>°</b> 1 395	56 <b>.</b> 88 2017	37,33 1324	31 <b>.</b> 47 1116	27 <b>.</b> 59 978	12 <b>,</b> 66 449	72 <b>.</b> 00 2556	1.23 44	10.87 385
9	14 <b>.</b> 72 179	8, 10 98	3 <b>.</b> 69 45	7 <b>.</b> 43 90	8 <b>,</b> 69 106	5,72 70	11.20 136	1 <b>.</b> 56 19	2,54 31
5 <b>I</b>	23 <b>.</b> 40 469	4 <b>.</b> 68 95	1 <b>.61</b> 33	10 <b>.</b> 04 201	5 <b>.</b> 09 102	4 <b>.</b> 78 97	4 <b>.</b> 12 82	1.14 23	3 <b>.</b> 02 61
	9 <b>.</b> 97 390	0.43 17	0.07 3	0.02 0.8	0.40 16	3 <b>.</b> 58 140	77.20 0.15 1775 6	0.07 3	0.07 3
-•	<b>15.</b> 65 360	67.41 1550	67,41 15 <b>6</b> 0	22,18 510	22 <b>.</b> 18 510	10 <b>.</b> 87 250	77.20 1775	9.13 210	14.13 325
9	7.9	7.8	8 <b>°</b> 5	7.50	1°1	<b>1.</b> 9	8°3	7.85	7.85
	3840	5060	4540	2750	2500	18 20	59 50	730	2130 1360
÷***	6000	1900	7100	4300	3900	28 50	0 <b>3</b> 00	1140	2130
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2	17. 19	50	51	22	23	24	25	26	27
• •••	17.	18, 20	19, 21	20. 22	21. 23	22, 24	23.	24, 26	26 <b>.</b> 27

( 3)

								0		6
	19		75.62 -	72.40 -	83 <b>-</b> 55 -	51 • 34 -	53•59 -	74,58 2,80	65 <b>,</b> 47 –	93,3810,20
	11		790	<b>4</b> 96	668	1162	655	242	580	235
	16		1.96 790	4.32	5 <b>•</b> 00 668	0.64 1162	3,32 655	5,00	0.80	7.64 235
	101 101		1.77 110	2 <b>.</b> 18 135	I	7,26 450	4 <b>,</b> 19 260	0 <b>.</b> 48 30	0.08 5	8 <b>.</b> 87 550
			10 <b>,</b> 68 652	8,82 538	7 <b>.</b> 94 484	7 <b>.</b> 35 448	7 <b>.</b> 74 472	7 <b>.</b> 64 466	7.78 475	14 <b>.</b> 90 909
			I	t	,	2	t	ı		
	12		8, 88 407	5 <b>.</b> 82 280	12 <b>.</b> 13 583	15,40 740	5 <b>.</b> 10 245	1.72 83	6 <b>,</b> 26 301	7.72 371
			44 <b>.</b> 41 1575	<b>19</b> .28 684	62,08 2200	33.07 1173	10 <b>.</b> 39 368	8 <b>.</b> 60 305	20.07 712	43, 45 1541
<u> </u>			10 <b>.</b> 97 133	4•63 56	7.48 91	14.14 172	8, 42 102	2.• 29 28	6 <b>.</b> 10 74	3 <b>.</b> 10 38
(4			4 <b>.</b> 84 97	5 <b>.</b> 30 108	5 <b>.</b> 88 120	9 <b>.</b> 10 180	4 <b>,</b> 68 95	2 <b>,</b> 55 52	5 <b>.</b> 50 110	1 <b>.</b> 60 32
			0 <b>.</b> 31 12	0 <b>.</b> 18 7	5,13 5	7 <b>.</b> 68 300	0 <b>•</b> 08 3	0 <b>.</b> 05 2	0 <b>.</b> 10 4	0,23 9
	· · · · ·		50,01 1150	26.53 610	68, 50 1575	<b>39,</b> 62 750	15, 22 350	14.35 350	22.18 510	69 <b>.</b> 59 1600
			7.75	7.6	7.5	7.8	8.05	7.55	8,1	7.95
	5		019E	2240	5180	3790	1670	1190	2180	4800
			6200	3500	8100	5900	2610	1860	3400	7500
		BHI NMAL BLOCK :	BAGAWAS	BHAGAL BHIM 3500	BHINMAL	DAMAN	DUMARIA	GOLIA	J ERAN	J ES AWAS
	1 2	BHI NMAL	26. 28	27. 29	28. 30	29, 33	<b>30.</b> 36	31. 37	32, 38	33. 39

	15 16 17 13 19	۰ بن ۱ - ۱		8 <b>1,95</b>	I 6	<b>;</b>	92 <b>.</b> 41 12 <b>.</b> 77	8	I D	49 <b>.</b> 64 1 <b>.</b> 08
	រ ខ្ម រ	74.55	<b>57.</b> 32	59 <b>°</b> 68	65 <b>.</b> 19	79.19	92.4	77 <b>.</b> 92	76.49	49 <b>°</b> 6
	17	1128	603	216	519	1615		665	1300	205
	16		196	2, 32	1.60	0,32 1615	14.20 268	1.96	5, 32	1.64 205
		1, 69 1, 69 105	0, 80 50	0.56 35	0 <b>.65</b> 40	2 <b>.</b> 90 180	0 <b>.</b> 80 50	8.87 550	3.06 190	0 <b>.</b> 08 5
	1 • 50 2 ====	10, 58 646	7.45 454	6 <b>.</b> 27 383	6. 47 39 5	10,40 640	14 <b>.</b> 99 915	11.74 700	11.27 688	5,19 317
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		12, 68	3 <b>.</b> 95 190	1.06 51	3 <b>.</b> 46 166	19.80 951	15.48 744	12 <b>.</b> 98 624	14.62 702	0.43
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~		7,90 24,66 101,50 12,68 10,58 160 300 3602 609 - 646	5 <b>.</b> 82 71	1.98 24	<b>6.</b> 05 74	21.80 265	4 <b>.</b> 00 49	9 <b>.</b> 56 116	18 <b>.</b> 30 22 3	1.72 21
<b>,</b> 2		7,90 160	4 <b>.</b> 37 88	2 <b>.</b> 34 48	4 <b>.</b> 32 86	10 <b>.50</b> 210	1 <b>.</b> 36 28	3 <b>.</b> 74 76	7.70 154	2 <b>.</b> 39 49
		0 10 4	1. 79 70	0.82 32	0.08 3.1	2.56 100	1.92 75	7.03 275	0 <b>.</b> 74 29	0 <b>.</b> 08 3
		93.70 2200	16,09 370	7.61	19.57 450	7.75 132.66 3050	88 <b>.67</b> 2139	71.76 1650	86 <b>.</b> 99 2000	4 <b>.</b> 13 95
		490 7 55	7,85	7.95	7,95	7,75	8,6	7.95	1.9	7.6
		7490	1730	780	2010	14100 9025	5760	5500	<b>67</b> 20	490
		11700 7	2700	1230	3150	14100	0006	8600	10 500	750
		U ETU	J UNJ ANT	KHUSALP URA	L EDARMER	M EDA	MISHOM	NANDI A	NAVARAUTA	NIMBAW2S
	 	<b>9</b>	41	42	43	38 <b>.</b> 44	45	40.46	47	48
	: <del>-</del> - 1	34	35	36.	37.	38	<b>3</b> 9 <b>.</b>	40 <b>.</b>	41.	42.

		c			6	7.	8.	9.	10.	11.	12.	13.	14.	15.	16.	17.	18.	19
	4.9. F	1. 2. 2. 1. 49. PHAGOTRA	3500	2240 8	2240 8.0	32.62	90 <sup>°</sup> 08	1.46 30	2.02 24	5	2.96 142		8,72	2.10 130	1.32 174	174	90.01 5.18	5, 18
	50 <b>.</b> I		1230	190	790 8.00	9.57	0,64 25	1.40 29			1.53	I	6.17 376	0.24	2.32 161	161	71.26 2.95	2 <b>.</b> 95
÷.	51 1	45. 51. RANGALA 1	10500	6720	۳. م ه	100.03 0.38 2300 15	0.38	4•00 80	3 <b>.</b> 00 36	864 81 3078	13.80 663	· •	16.09 3.39 982 210	2.33	7.29	350	93.13 9.09	60 • 6
2	52.	46. 52. SEWRI	0079	7090		<b>35.88</b> 825	0.15 6	18.20 365	12.60 153	55.48 1967	6,52 313	L	5.18 316	lil	Lin	1540	53.69	1
·**	53.	T. 53. SOBRANAS	6100	3900	7.85	56.04 1300	0,08	0•94 19	2 <b>.3</b> 9 29	36.29 1287	6.34 305	I	13 <b>.</b> 72 837	13.72 4.19 837 260	8.00	166	94.31	10.39
-																		
48. 48.	56.	48. <u>JALORE BLOCK</u> : 48. 56. BAGRA 450	4 500	2880	7.75	34.80 0.66 800 26	0.66 26	7.24	4•94 60	32.73	3.39	ı	10.94 668	10•94 0•40 668 25	1.56	609	73.05	ĩ
•6*	57.	ROAD	3900	2490	7.7	34•14 785	0.04 1.4	34.14 0.04 1.73 785 1.4 34	3.26	19.39 688	7.25 348	I	11.45 699	0, 32 20	450	250	87.16 6.46	6.46
50 <b>.</b>	59.	50, 59. BIRBAR	54:00	3950	7.45		3.5	<b>39.15 0.09 4.60</b> 900 <b>3.5</b> 92	9.60 117	40.61 1440	3.99 192	· 1	8 <b>.</b> 90 540	1.05 65	3.60 710	710	73.26	1

10460 7.35 102.21 0.33 66.50 21.10 165.73 9.80 56.50 22 5.45
33.66 0.07 5.04 5.82 29.38 4.63 10.38 0.24 5.64 543 75.46
2,52 0,12 2,49 2,50 1,79 0,49 7 5,09 10,40 1,249 33,03
2.39, 3.32, 4.71, 2.64 2.88 0.63, 5.594 3.31 5.40 1.3368 1.18.30
6.08 0.17 6.34 5.72 5.67 2.32. 5.68 64.59 5.432 5.60 7.33.20
47.84.0.12 2.39 4.89 37.98 7.29 1.2.0 2.0 2.0 2.0 2.0 2.0 2.2 2
59. 81. 0. 10 16. 84. 19. 04. 85. 98 3. 88. 2. 6. 07 7. 9. 65 19. 0 5 19. 0 5 19. 2. 69. 65.
11. 96. 1.02 1.14. 2.13 2.64 1.111 2.111 2.10156 2. 8.00 4. 163. 18.60

	<b>.</b> .	4.	<b>5</b>	6.	7.	•		10.	11,	12, 1	13. 14.	15.	10.	-11-	19.	19.
H4	60. 76. REMAL	5700	3650	8 <b>.</b> 0	27.83 640	4.34	8.74 175	12.63	41.49 1473	4•48 215	4•80 293	0 2.50	i v	€3 \.'.' ∓=	51.93	1
-	61. 71. SITANA	1320	845	7.25	8.70 200	0.46 2 18	2.25	1.28 16	6.14 218	1+22 - 59 -	5.30 323	0.73	2.80	177	68.56	1.77
21	455 MANTFULA: 62. 72. BHARUDI	1290	151	7.6	8.70 0 200	0• :4 1•6	1•92 39	<b>2.1</b> 6 26	4•12 146	0.06 3	7.95 485	5 0.40 25	2,00	204	67.86	3.87
	63. 73. BUSLON	5850	3520	7.6	48•71 0 1120 4	0.11 4.4	3 <b>.</b> 36 67	4.42	42°14 1496	11.21 538	5.72	2 0.73	4.40	389	86 <b>.</b> 06	I
	64. 74. CHANDUR	1275	738	7.65	6.31 145	3.1	3.65 73	2.49 30	5,20 187	0.69 - 33	7.31 446	i 	0\$0	307	50,36	1.17
	65. 75. DHANJI-KA- BÂÑA.	3950	2170	7.4	20,10 0.06 690 2.3	0.06 ; 2.3	3.35	8,32 107	23°CJ 994	2.24 - 106	7.31 446	0.57 35	3.60	609	68.08	1
	66. 76. DHANSA	945	583	7.6	6.00 0 138 1	0.04 1 1.6	33	1.87 23	3. 19	0.26 - 12	6.78 414	5 0.24 15	1.60	178	62.56	3.23
	67. 77. GOLANA	2115	1150	1.1	11.41 263	0.15 3 6	3.66	4.71 1 57	10.67 378	2.08 100	7.62 465	t	2.80	419	57.25	8
	68. 78. JHAK	5750	3520	7•4	44.36	1.07 5.18 42 104		6.82 3 83	36 <b>-</b> 48 1293	7.36 <b>-</b> 354 -	10 <b>.</b> 39 634	9 4•04 250	<b>5</b> <b>5</b>	600	77.24	r

	2.	3.	4.	5.	¢•	7.	ಕು	.6	10.	11	12.	13.	14.	15.	16.	17.	16.	19.
69.	79.	69. 79. JASWANTPURA 1125	4 1125	625	7.45	5.44	0.06	2.88 58	2.78 34	4-51	0.33	3	6.15 375	0.08 5	2.40	283	48.74	0.49
70. 1	80	70. 80. KEBIA	3700	2240	56-1	27.84 640	r L	3.43	5.51	18, 96 670	4•06 195	ı	12.25 747	1.21	00 <b>•</b> †	144	75.69	3.31
71.		71. 81. KORITA	4000	2430	7.6	32.62	5.1	4.51	4.12	23.81 844	5.24	ť	11 <b>.</b> 02 672	0.48 30	0,40	432	78.83	2.39
72.	82.1	72. 82. MANDOLI	2640	1510	7.4	19°14 440	0 <b>.</b> 16 6.1	3.07	2.88 35	15.55 551	1•61 77	T	7.84 478	0°.97	2•00	298	75.80	1.89
13.	83.	73. 83. MODRAN	1950	1150	6-1	8. 70 200	0.03	5.33	6.38 78	12 <b>.</b> 00 425	6 <b>.</b> 05 149	1	4•66 284	0.40	0**0	586	42.56	ı
74.	84.	74. 84. MORTARA SILLI	4550	2910	7.45	24: 36	<b>3.</b> 99 156	9, 4,6 190	9, 55 116	17.44 678	12 <b>.</b> 68 609	1	8.37 511	9•68 600	4.80	951	51.44	1
75.	85 <b>.</b>	75. 85. PANSERI	430	270	7.85	0,60 14	7r.	2.65 54	1.35 16	0.47	1		3.82 239	0.40	<b>0.6</b> 4	200	13, 04	ı
76.	86.	194 •98 •97	1335	760	7.15	5.44	0.14	4•03 80	3.55	4•41 156	0.92 44	• 1	7.21	0.65 40	0.80	379	44.33	· •
777. 88. IAT OL	23 23	TAT OL	5050	3000	7.3	30,88 710	0.56 22	9 <b>.</b> 98 200	8.64 105	37.64 1335	6.68 321	1	5•94 362	1.62	0.80	931	61 . 68	ŧ

1, 2,	3.	4.	2	6.	7.	చ	<b>9.</b>	10.	1.	12.	13.	14.	15.	16.	17.	18:	19.
•	RANIWARA BLOCK	1. DCK									ł						
78, 89,	78. 89. 4JODAR	079	410	7•6	2.74 63	0.17	2.91 59	1.46 18	1.93 70	0 <b>.</b> 33 16	1	4,60 293	ŧ	1,00	218	38.16	C <b>†</b> *0
79, 90, ALRI	ALRI	1950	1240	\$•¢	14•44 332	0.13 <sup>5</sup>	2.60 53	3.85	9.73 345	<b>6</b> 67 67 7 7 7 7	0.73 23	5•70 353	0	1. 96	322	60, 80	<b>0.</b> 4
50 <b>.</b> 91.	50. 91. BAMANWARA	960	614	7.7	4.67	6.64 33	3.17 65	2,29 20	1.70 60	0,07 3	•	9.41	· 1	0.64	273	43.60	3.95
ΰ <b>1. 92.</b>	81. 92. BARETHA	2760	1760	ů <b>.</b> 05	11.30 260	0.07 3	4+95	7.19	10 <b>.</b> 23 364		r	7•55 461	2.10 130 4	4.40	606	48.10	1
32 <b>-</b> 23	CZ. TS. CHATWARA	5350	3420	7•0	44.50 1025	0,36 14	2.39 49	6.66 81	36.27	5.32	I	6 <b>.</b> 04 491	2.26	3.64	452	C2 <b>.</b> 57	I
63. 94. DADOKI	DÅD0 KI	5100	3260	7.75	30,05 875	0°74 29	8, 37 168	40.05 61	27°12 962	7.61 364	7	10. 19 622	6•61 2 410 2	2.96	668	72.96	
84. 95. DHAN OL	TO NYHO	1470	076	7. 65	6.08 140	2.43	3,02 62	3.27	4•63 164	53	1	6•37 369		0°64	314	41.08	0 <b>•</b> 0
85, 96.	85. 96. DHANWARA	1360	0.00	са. Р	<b>19.</b> 00 230	0.66 26	2.03 41		4.62	1.73	<b>F</b> '	7.64 466	0.64 3	3.64	202	96 - 29	3.59
86. 97. DIG40N	DIGAON	5500	3520	<b>2</b>	52.19 1200	0.03	1.09	2.96 36	39.59	2, 13 102	1.96 59	11.56 705	0.72 4	4.64	202	92.67	7.5

19.		4•64	T	2,51	1	1.51	1	ł	•
18.	33.13	73.98	38.07	48.50	19.09	43.87	52,85	57.63	41 <b>.90 1.</b> 06
17.	321	189	793	197	568	218	362	1386	270
16.	0.80	3.32	0.64	0-80	0 <b>°</b> 64	1, 96	2.32	0.32	1, 64
1.7	0.56	09 60	0.24	0.72	4• 92 305	0.24	0.72 45	1	· E
14.	5.87 358	8.43 514	5.19	347	4.80 293	5.88 359	6•47 395	12.25	6.17 395
:5:		· •	· 1	0.78 23	:	r	I	1.37	· T
12.	0.52	2.77	132 87	2.31	1.45 70	0.28	3.02 145	3.94 189	0.23
11.	2.40 05	5.38	17.86 633	<b>2.3</b> 6 84	4.82 171	1.32 47	20.22 717	48,38 1715	2.55 90
10.	3.85 47	2.07	5.88	2.44 30	4.53	1.93 23	9.21	17.53 213	2.24
9.	2.55 51	1.72 35	9,98 200	<b>1.51</b> 31	6.86 137	2.44	5.56 111	10.19 204	3.17
<b>в</b>	0.05 2	0.95 37	0,05 2	2.05 BU	2.43 95	0.07 3	0.28 11	0, 25 10	0 <b>.</b> 08
7.	3.21	13.48 310	9.78 225	5.45 130	3.26	3.47	15.87 388	38.04 875	3.96 91
<b>6.</b>	21.75	β <b>.</b> 00	7.35	3.4	7.7	7.75	7.8	8.4	7.65
5.	535	1060	1650	670	1070	450	1920	3900	530
4.	810	1665	2580	1050	1680	004	3000	6100	830
3.	87. 98. FATEHPURA MALWARA	GÅ NG	89, 100. GOLHARA	90. 101. HARSHWARA	91. 102. JETPURA	92.103. KAGMAL	Kårdå	94.105. KORKA	95. 106. LAKHANAS
°7	98,	88. 99. GANG	100.	101.	102.	103.	93.104. KARDA	105.	106.
	87	88	8 <b>9</b>	<b>06</b>	91.	92	93.	94•	95.

1, 2,	3.	4.	5	9.	- <u>7</u> -	сэ <b>.</b>	<b>9</b> .	8	11.	12.	13.	14.	15.	16.	17.	18,	19.
96.107.	96.107. MARDWLRA	515	585	6-1	4.43 102	0.18	2.24 46	3.53	2.93 104	0.69 33	1	5.68 347	0.72 45	1.64	2 83	42.68	
97. 108. PAL	PÅL	1125	720	7.7	65 - 83	4.35	2.34	2, 29	1.79 63	1.30 62	٠	5.49	2.58 160	1.00	231	23,96	0.86
98 <b>.</b> 109.	98.109. RAMPURA		275	7.6	0.78 18	0.12	2•44 50	1.30	0.57 20	•	t	3. 82 233	0.40	0,64	187	16.81	0.08
99.110.	99.110. RANIWARA	1140	730	1.1	3.04 135	0.54 21	2.08 42	2.13 26	3.31	0.23	ı	9. 31 568	0. 24 15	1.64	210	6 <b>2.</b> 86	5.10
100.111	100.111.RATANPURA	1020	650	7.45	5.74 132	0.18	2.29 47	3.69 45	4•06 144	0,90 43	1	5, 88 359	0.24	1. 32	299	48, 23	ŧ
101.112. ROPST	ROPST	5750	3680 7.6	7.6	43° 50 1600	0,33 <b>13</b>	8, 01 160	7.75 94	45°93 1620	7°08 340	1	5.68 347	0.64 40	3.32	788	73.00	• 1
102.113. R. R.	• RORA	1140	730	7.55	3•26 75	3.32 2	2.24	2.55 32	2.17 77	0.80 38	1	7.05 430	75	1.96	244	28.42	2.16
103.,114.	103.114. SEWARA	44c0	2810	7.45	27.40 630	0.15 9 6	200	8 <b>.</b> 06 98	25 <b>.</b> 99 922	5 <b>,</b> 20 250	1	5 <b>.</b> 09 311	8.06 500	0.32	, 889	60,17	ı
104-115. VANDAR	VANDAR	1170	750	7.85	6.96 160	0.07 3	3.43	2.76 34	2.55 90	1, 16 56	I	7.55 461	1.29 80	1.00	309	52.65	1.36
105. 116.	SANCHORE BLOCK: 105.116. AGHALPURA 5200	<u>BLOCK</u> : 5aCC	3330	ŝ. 15	36.97 8 <b>5</b> 0	0.35 4. 14	4.93 1	12. 66 1364	38.46 1364	7063 366	1	6.96 425	1.37 85	1. 12	879	67 <u>.</u> 33	I

1. 2.	3.	•+	~	6.	7. B	8, 9	9. 1	10.	11. 15	12. 13.	14.	15.	+	17.	18.	19
106.117. AGDANA	AG DA VA	3650	2330	7.7	13.04 0.07 300 3		\$ 29 1 166	4 01 2 998	8.29 14.01 28.16 1.44 166 998 998 69	• 6	4•99 299	1.00	4•00	1 1 1 1	<b>36</b> . 82	
107.118. AMLI	AMLI	13650	8730 7.75	7.75	137.01 5.11 8 3150 200 1	5.11 8 200 1	8.74 175	13.78 163	116.94 4146	13.78 116.94 1.76 - 4 163 4146 84	42.60 2599	0 <b>.</b> 46	4.00	1126 <sup>84</sup> .33	4.33	20.08
108 119. ANKHOL	ANKHOL	19200	12290 7.65		173.98 0.87 14.95 3 4000 34 299	0.87 1 34	14•95 299	<b>38.</b> 60 469	189.45 6717	21.90 - 1052	7.25	5•64 350	3.32	2677	2677 76.17	F
109, 123,	109, 123, BHADRUNA	3400	2170	7.9	20.01	6.14 2.62 240 53	: 62 53	4.83	16 <b>.</b> 73 593	2.70 - 130	9.51 580	3.63	0.32	372	59.55 2.06	<b>2.</b> 06
110.124.	110.124. BHAWATRA	0066	6330	7.5	6 <b>3.2</b> 8 1455	0.69 21.52 1 27 431	1.52	19.12 232	91.00 3230	6.14 <b>-</b> 295	4.21 257	1	,	2032	60•49	F
111.125.	111.125. BINJR OL	2840	1820	L * L	15•65 360	10.23 1.05 1.02 400 21 17	1.05 21	1.02	12.09 429	2.91 - 140	<b>11.</b> 76 718	1.12	7.32	123	55,20	9.29
112, 127. BIRAWA	.'	4100	2620	7.3	21-31 490	1.02 40	9,80 196	9,80 9,14 196 115	89.48 1045	3.19	3, 88 237	4•92 305	0*0	962	51.25	
113, 128,	113.128. CHITALMANA 3900	3900	2490	7.3	29 <b>.</b> 57 680	0.25	5.00	4.89	5.25 4.89 24.29 117 59 861	6 <b>.</b> 27 _ 301	7.45 455	1.61 100	I	512	73.83	I
114.130. DEDUA	DEDUA	6000	3840	7.8	40 <b>•</b> 23 925	3.19	5.35 109	14°18 172	5.35 14.18 46.97 109 172 1665	5.01 241	7.55	2.82 175	1.32	976	6 <b>3.</b> 91	8

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L, 2.	с.	4.	6	5	•	•	••	j			5	ł			•		
115. 131. DHANI- GUDA HEMA	DHANT- GUDA HEMA	11400	7290	7.25	75•02 1725	6 <b>.</b> 39 250	6.20 124	12•43 152	71 <b>•44</b> 2533	11.48 551	1	9 <b>.41</b> 1.77 574 110		25 • •	934	75•48	ĩ
116. 132. DHANTA	DHLNTA	3400	2170	7.75	17 <b>.</b> 83 410	4.35 170	7.12 4.99 143 61		15•95 566	3 <b>.</b> 29 158	I	5,51 8, 330 52	8•33 520	3	605	00.5	3
117. 133. DUGAWA	DUGAWA	4900	3130	7.65	23.92 550	10.23 400	7.66 9 153	9.93 121	22 <b>.</b> 96 814	12.41 596	3	11.86 5 724 3	5•24 325	1	899	46.23	r
118. 134. DUNGRI	DUNGRI	4900	3130	7.7	23•80 0•25 547 10	0.25 10	9 <b>.</b> 76 196	18.64 5 227	28 <b>.1</b> 6	11.62 558	t	6•08 371 3	5•70 <sup>1</sup> 350 <sup>1</sup>	1•32	1420	45,38	B
119. 135. GARDALI	GARDALI	6100	3900	7.85	25,09 600	15•35 600	5.76 115	8.28 101	39∙50 1400	7 <b>•</b> 23 347	3	7•55 <sup>]</sup> 461	1.21	2.60	702	47.03	ŧ
120. 137.	120. 137. HOTIGAON	4600	2940	7.85	32,05 375	0•35 14	3.67 73	5°30 72	52.51 1153	6,47 311	t	7.35 448		0.64	477	79.35	F
121. 139. JHAB	JHAB	9270	5890	7.75	70 <b>.</b> 67 1625	6 <b>.</b> 39 250	6•20 124	12,43 152	71.44 1533	11.48 551	1	9.41 1. 574 1	1.77 3 110 3	3.32	934	73•81	t
122. 140.	122. 140. JODHAWAS	3650	2330	7.55	26 <b>.</b> 09 600	1.02 40	6.19 124	4.20 51	23 <b>.</b> 81 844	4 <b>•</b> 75 228	ı	6.76 0. 412	0 <b>.</b> 72 45		519	69,57	I
123. 143. KHEJDI- YALI	KHEJDI- YALI	9600	5760	7.65	64.93 1495	1 <b>•91</b> 75	15 <b>.</b> 96 320	12.92	78•06 2 <sup>768</sup>	9 <b>•3</b> 8 451	ŧ	5 <b>.</b> 59 341		1	1444	67.83	

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-16 16		1	ł	ł	۲	ı	1	r <b>s</b>	6.40	1
13.	36.44	76.57	63 . ၂၇	53•78	67.73	67.11	58•96	1 <b>8°</b> 64	86,40	75.75
<u>1</u> 7.	15 <b>3</b> 8	1512	1743	307	622	1055	953	871	159	1438
<u>ୁ</u> ମ	1.64	0.64	2.64	1.32	10.19 0.24 622 15 1.64	12•94 7•26 790 450 1•32	1.45 90 0.32	2•66 165 2•32	0•48 30 <sup>8</sup> 8•00	2•42 150 1•64
15.	10	E	1•85 115	1.69 105	) 0•2 15	: 7•2( 450	н 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2•6( 165	0•4£ 30	2•42 150
14.	11.07 0.16 675 10	<b>4</b> •70 287	4 <b>.</b> 80 293	5 <b>.</b> 68 347	10 <b>.19</b> 622	12 <b>.</b> 94 790	8•33 508	5•68 347	9 <b>.</b> 59 585	6 <b>.</b> 08 371
13.		1 92	ı O	1	1	3	1	-	1	•
R.	2.73 131	6 17 <b>.</b> 36 834	2 11•06 5 531	3•01 145	<b>19.56 8.</b> 05 694 387	: 6 <b>,</b> 76 325	- 6.43 309	10•57 508	2•67 128	97•14 12•84 3444 617
11.	45 <b>.</b> 03 1598	104.7( 3712	2795	9 <b>.</b> 35 332	<b>19.56</b> 694	58 . 73 2084	34.30 6.43 1216 309	68 <b>•</b> 80 2439	12.44 441	97•14 3444
10.	18.16 221	7.98 22.26 104.76 160 271 3712	16.16 18.70 323 227	<b>3</b> •52 43	8.19 100	15.32 136	10.35 126	8 <b>.</b> 93 108	2.26 27	11.76 17.00 236 207
6	12.60 18.16 45.03 2.73 253 221 1598 131		16 <b>.</b> 16	2•62 53	4 <b>•</b> 25 85	5•78 : 118	8.71 : 174	8•50 170	0.93 19	11•76 236
8.	7.16 280	: 0•36 14	0.17 7	3.20 125	0•20 8	7•67 300	2•3) 90	0.33 13	0 <b>•3</b> 0 12	0•48 19
7.	21.74 500	100•03 0•36 2300 14	61.98 1425	10 <b>.</b> 37 250	26 <b>.</b> 53 610	58 <b>.</b> 71 1350	29 <b>•</b> 57 680	70.67 1625	22 <b>.</b> 18 510	91 <b>.3</b> 4 2100
9	7.4	8•15	7.65	7.75	8.10	7.75	7.75	7.60	8•05	7.50
5.	3770	7490	5950	1300	2240	5310	3130	5500	1750	00TL
4.	5900	11700	9300	2040	3500	8300			2730	00111
	124. 144. LALPUR	125. 145. MELAWAS	126. 146. NALDHRA	127. 147. NIMBAU	128, 148, PATHMEDA	129. 149. PLEDAR	130. 150. RMPURA 4900	131. 151. RANODAR- 8600 KA-GULLA	132. 152. SAKRIA- KI-DHANT	
·0	144.	145.	146.	147.	143 <b>.</b> F	149.	20	21.	52.	4.SA1
	124.	125.	126.	127.	128, ]	129. 1	130. 1	131. 1	132. 1	133.154.SANGDAVA

C.	14	1	6.27	t	3	,		ı	8.76	•
	•	67 <b>.</b> 61	80.75	17.°	33.74	76.82	86 <b>•</b> 63	86.51	74.08	49.30
17	• •	945	372	1790	265	561	367	456	225	1279
16.	5	0.64	7° 64		0•64	<b>1</b> •6₄	3.64	3.32	7.64	0.32
15.	1	2•42 150	200 200 200	ł	0.24 15	3 <b>.</b> 63 225	<b>4.51</b> 280	0;16 10	3 <b>.63</b> 225	0.56 35
14.		9•55 461	13 <b>.</b> 72 837	5•90 360	.5•29 317	<b>5.</b> 38 359	5•09 311	5.49 335	15.23 805	7•05 430
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61		6 <b>.</b> 48 311	9 <b>.</b> 62 ⊈62	10 <b>•</b> 20 490	0 <b>.</b> 85 41	7 <b>.32</b> 352	6.48 311	5,62 271	5,02 289	1.06 51
11	i	41 <b>.</b> 86 1484	32 <b>.</b> 98 1161	<b>82.</b> 22 2915	1•42 50	42 <b>.</b> 81 151	39,50 1400	<b>54,15</b> 1928	36,82 1305	42•52 1508
-01	5	9 <b>.</b> 87 120	4.72 57	18 <b>•38</b> 224	2•52 31	6 <b>.3</b> 5 77	5•51 67	5 <b>.</b> 14 63	2•83 34	12.90 157
6	,	9•03 181	2•73 55	17 <b>.</b> 22 345	2•78 57	4 <b>.</b> 88 98	<b>1.</b> 84 37	3 <b>.</b> 99 80	1•67 33	12•68 254
Ø	5	0 <b>.</b> 36 14	4.47 175	2•55 100	0•08 3	2•55 100	0 <b>•20</b> 3	0 <b>.</b> 17 7	7 <b>.</b> 67 300	0 <b>•10</b>
7.	:	40•23 ( 925	47 <b>.</b> 34 110	56 <b>•</b> 54 1300	2.74 63	45•67 1050	48 <b>.</b> 93 1125	59 <b>.</b> 30 1375	34.79 800	25 <b>.</b> 00 575
9	;	7.8	8•00	C•2	7.45	7.75	8,25	3.30	8•05	7.3
e.	5	3580	3650	5890	416	3370	3580	4290	3340	3330
4.		5600	5700	9200	650	6050	5600	6700	6000	5200
e	-	ARNAU	135. 158. SIWARA	UNIR	URAWA	Idm	139. 162. TENTROL 5600	ANK	IROL	SATLA BLOCK: BAKRA 5200
		134. 156. SARVAU	158• 8	136, 159, SWIR	137. 16). SURAMA	138. 161. TAMPI	162 <b>.</b> 1	140. 163. VANK	141. 164. WIROL	142. 165. BAKRA
	;	134.	135.	136 <b>.</b>	137.	138.	139.	140.	141.	142.

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166.	143. 166. BISALÀ	5600	3530	7.7	<b>34.7</b> 9 800	0.35 14	10.92 219	6•50 79	38•08 1350	10•21 490	•	6 <b>.37</b> 339	0°36 60	1	175	- 6I.9 <b>9</b>	t t
167.	144. 167. BLUTR.	2530	1650	7.4	7 <b>.</b> 82 180	0•15 6	7.28 1/29	7.12 87	9•07 322	3.55 170	ı	4.61 231	6, 32 375		720	35•00 •	t
168.	145. 168. BISAN- GARH	10200	6530	2.6	69 <b>.</b> 59 1600	0•20 8	19 <b>.</b> 34 388	20•23 247	87.50 3102	10.78 518	T	9•30 598	0•40 25	0•40	1361	63.60 .	1
169	146. 169. CHARAU	1290	825	7.65	6 <b>,</b> 52 150	0 <b>.1</b> 5 6	1•92 39	5•00 61	5 <b>.</b> 20 184	2 <b>.</b> 98 143	ı	6 <b>.</b> 08 371	0.56 35	1.61	316	48,00	E
170.	147. 170. DADEAL	5050	3230	8•00	48.93 1125	0 <b>.</b> 69 27	4 <b>.</b> 26 85	3•70 45	27 <b>.</b> 50 975	6.45 310	ł	22•05 1345	0.16 10	6 <b>.</b> 32	398	85•01 1⁄	l 14•09
171.	148. 171. DAHIWA	2769	1765	7.35	20,88 480	0.25 10	4 <b>.</b> 21 84	2•81 84	15•78 559	3.74 180	T	6 <b>.</b> 96 425	0.89 55	1•00 351	351	74.17	ı
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150. 173. GOL	ŢOĐ	1890	1210	7.9	9 <b>.13</b> 210	2.81 110	3•33 68	2•65 32	6 <b>.</b> 52 231	2•81 135	E	3•23 502	0•48 30	1.00 299		50.94 2.25	15
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	11.	8.16 239	34.96 1240	4.54 161	16 <b>.</b> 25 576	35•7⁄± 1267	5.67 201	48 <b>.</b> 67 1726	1•51 54	116 <b>.</b> 20 4120	50.37 1786
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	9.	0.41 8	<b>1</b> •40 28	<b>1.25</b> 26	6 <b>.</b> 50 130	3•03 62	6•40 128	6 <b>.</b> 45 129	<b>1.</b> 92. <b>39</b>	13.50	5 9 <b>.</b> 88 198
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	2.	22•61 520	8•00 <sup>48</sup> •93 1125	9 <b>•1</b> 3 210	21 •31 490	46 <b>.</b> 75 1075	6•30 ∎145	59 <b>.</b> 80 1375	4.48 103	119.61 2750	46 <b>.</b> 75 1075
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	e e	ARMU	153. 176. KHATLAUA	154. 177. KESHWANA	155. 178. KUABER	156. 179. MEGALWA	157, 180, MOKNI- KHERA	158. 181. OTHWARA	159. 182. PAHAR- PURA	IGO. 1831 PANTHERI	161. 184. Ratunja
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1%. 1	3•80 183	4 <b>.68</b> 225	7 <b>.</b> 81 375	1 <b>.</b> 56 75
11,	27•32 720	2:•19 3 <b>5</b> 3	54.71 1940	5 <b>.</b> 76 204
10.	6.45 78	10 <b>.</b> 76 131	8,64 105	2•29 23
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celo	1•27 50	0•10 4	0 <b>.56</b> 22	0•05 2
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చ	7.75	7.65	7.5	7.9
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ير مرجع م	17. 18.	90.15 5.17		73 •60	1 31	81 <b>.</b> 04	56.07 0.45	50.62 +	63 52 -	63 .76 -	- Well was abondoned saline formation.
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Æ	è.	8.75	<b>3</b> •6	7•80	7 <b>.</b> 8	7.95	7.5	8 <b>.</b> 05	7.95	1.1	7.15
je j	-4 -4	958	20 <b>60</b>	3300	2180	5580	550	574	6930	<b>9</b> 720	1
ECX10 <sup>°</sup> (Mi <i>ero</i> -	Siemens/cm (mg/t 3. 4.	009T	3400 1)	5555	3600	0068	880	930	14 <b>,</b> 100	112 M. 15,300	35,210
S. LOCATION	2.	1. A Luk RA		BAGRA	BAGRA ( Dos. )	BAUTRA	BAUTRA 880 (Test well)	BAUTBA	ealliara 42 m.	112 M <b>.</b>	EHADRAJUN 35,210
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3.	5650	1470	3000	5800	6200	R 5000	6200	6600	1080	1305	480	. 450	
2.	JODHAMAS 5650 3358	25. KOLAR	CUABER 50 M.	KUABER	L (U N. KORATA	23. KHAN PUR 5000 3100	<b>TEDA</b>	30. MEDA (0.5.s)	MODRAN	32. MODRAN	PANSERI 480 (Obs)	34. PANSERI 450 (0bs)	
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16.		(~1 2112	582	577	567	525	1076	243	333	343	243	5880
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13.	6•≙9 396	6•03 368	9•92 605	9•69 591	9•53 585	13•08 798	12 <b>.</b> 34 753	7.78	8 <b>.</b> 63 527	8•34 539	7 <b>.</b> 78 475	3.72 227
12.	ı	s	Ł	ŧ		ı	I	L	ı	1	ľ	1
11.	0 <b>°</b> 96 46	0•75 36	5 <b>.</b> 97 237	6. 04 290	5 <b>.</b> 36 231	4.56 219	15.49 744	1•33 68	2 <b>.33</b> 112	2•83 130	1.43 69	144.10 6920
10	6•47 229	6.57 233	63 <b>.</b> 77 2264	63 <b>.</b> 46 2253	63 <b>.</b> 73 2260	51•22 1783	52•73 1372	11.66 413	15 <b>,</b> 98 567	16 <b>.63</b> 590	12.03 427	389.30 13300
6	1•20 15	1.20 15	7.39 96	8•06 98	8•09 98	6.41 73	17 <b>.</b> 93 218	2•90 35	3 <b>.</b> 35 41	3•45 42	3•05 37	94•00 1143
ŝ	1.20 24		3•74 75		3•24 65	4•04 81	3•49 70	1•95 39	3•30 66	3•40 68	1.80 36	23•60 472
7.	20 10 10 10	•05	3 0.08	3°08	0•08 3	0•08 3	0•10 ∉	0•03 1	0•10 4	<b>0•1</b> 0	0°0	۲ 8 8
6.	11.74 270	11.74 270	68.51 1575	68.51 1575	68 <b>.</b> 51 1575	56 <b>.</b> 43 1299	59•74 1374	16•09 370	19•60 0•10 450 4	19•60 ( 450	16 <sub>6</sub> 09	435,00,20 435,00,20 10000 8
5.	7,95	7,95	0•0 8	7.95	7.95	7.⊈	7.0	<b>3</b> •05	7.65	7.7	3 <b>•</b> 15	8 •2
4.	830	800	4800	4820	4720	4090	4320	1190	1500	1600	1220	32490
3.	1320	1320	1 7680	1590	▲7700	1 6400 -29)	6750	2036	2596 rd	2711 th	2037	35000
	4		45, 81 MKA RANA 7680	SANKARANA 7590	SLATCARANA #77	46. SANKA RANA 6400 (0bs)(24-29)	SANKARANA 675 (17-22) (24-33) (38-49)	47, UMADPURA 2036 - IInd	UMADPURA 2596 StageIIIzd	UNEDPURA 2711 Stage IVth	48. UNEDPUTA 2037	19. TARKARA 35000
	4		45°	, ⊶£00 ,	67	46 <b>.</b>		47 <b>,</b> ⊓	പറു	10	58°.1	49 <b>•</b> I

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<b>%</b>	36. POALI	1750	) 1160	8.0							<b>،</b>	723	75	<b>2•</b> CO	190	RO* ZA	3•0 <b>5</b>
N	ar. PUNASA	1410	860	8 <b>.</b> 2	320 19.87						1	5•66 345	<b>6.6]</b> 410	7•0	1:55	33-70	2,56
R	PUNASA 243 M.	1590	1010	7.9	250 13 <b>•</b> 91		30 0 <b>.</b> 97	21 0.76			:	6.47 395 6.07	07 <b>0</b> 30	1.32	160	75.70	3.26
•	PUNASA M. 711	1530	930	0 <b>•</b> 8	320 13,04		-	6 7		125	•	389	0•40 25	<b>3</b> •00	36	83.9	4.6 <u>4</u>
8	RAJANMADI (Obs)	1650	1050	8,25			18 0.83	12	ດ ເຊິ່ງ ເຊິ່ງ	123		6 <b>.</b> 14 375	0•03 5	3.36	97	35•45	4.20
	39. RAJANNADI	1890	0611	7.8	384 13 <b>.</b> 65	0 0 0	1.10	6 6	150	1•09 52	1.06 31	11.39 695	0.16 10	7.0	22	92.05	11-05
å	ED. RAITHAL	43000	37240	7.55	429 478 4	1 5 .33		70.02	228 228	1•29 62	t	12 <b>.</b> 34 753	ач.	6 <b>.</b> 3	86	69 <b>•</b> [6	10.67
<b>4</b> 1. RAM	AMA	3796	2190	7.5	11000 13	61 61 61			4°8•40 16628	) 135 <b>•9</b> 0 = 6524	I Q	8,93 543	1	10•00	5020	82.61	,
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_ A <b>4</b> ***	E. RANANARA 28 M.	1080	700	8 <b>•</b> 05	500 6.96	3 0•15	2.00 2.00	83		1. 68 31 31	1	5.11 312	1•29 80	1	791	57.79	+
AR C	RANTWARA	1080		0.8						C 18	1			1.6	221	6),36	3.15
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